

(19)



Europäisches Patentamt  
European Patent Office  
Office européen des brevets



(11)

EP 0 794 253 A2

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication:  
10.09.1997 Bulletin 1997/37

(51) Int. Cl.<sup>6</sup>: C12N 15/57, C12N 9/48,  
C12N 9/50, C12P 21/06

(21) Application number: 97103862.5

(22) Date of filing: 07.03.1997

(84) Designated Contracting States:  
CH DE FR GB IT LI

(30) Priority: 08.03.1996 JP 51848/96  
14.02.1997 JP 30458/97

(71) Applicant: Ajinomoto Co., Ltd.  
Tokyo (JP)

(72) Inventors:  
• Asano, Minao,  
c/o Ajinomoto Co., Inc.  
Kawasaki-shi, Kanagawa-ken (JP)  
• Kawai, Misako,  
c/o Ajinomoto Co., Inc.  
Kawasaki-shi, Kanagawa-ken (JP)

• Miwa, Tetsuya,  
c/o Ajinomoto Co., Inc.  
Kawasaki-shi, Kanagawa-ken (JP)  
• Nio, Noriki,  
c/o Ajinomoto Co., Inc.  
Kawasaki-shi, Kanagawa-ken (JP)

(74) Representative: Kolb, Helga, Dr. Dipl.-Chem. et al  
Hoffmann Eitle,  
Patent- und Rechtsanwälte,  
Arabellastrasse 4  
81925 München (DE)

Remarks:

The applicant has subsequently filed a sequence  
listing and declared, that it includes no new matter.

(54) **Aminopeptidase GX, and a method of hydrolyzing a protein with the same**

(57) The object of the present invention is to provide  
aminopeptidase efficiently decomposing a low-molecu-  
lar-weight peptide containing glutamic acid or aspartic  
acid in its sequence as well as a method of hydrolyzing  
a peptide or protein by use of said aminopeptidase.

Aminopeptidase GX being derived from germinated  
soybean cotyledons and releasing glutamic acid or  
aspartic acid from a peptide or protein containing  
glutamic acid or aspartic acid at the N-terminal, and a  
method of hydrolyzing a peptide or protein by use of the  
same.

EP 0 794 253 A2

## Description

The present invention relates to a novel aminopeptidase and a method of hydrolyzing a peptide or protein by use of said aminopeptidase or a cell extract containing said aminopeptidase.

## [Prior Art]

The aminopeptidase is an enzyme which releases an N-terminal amino acid sequentially from a peptide or protein.

The decomposition of proteins such as soybean protein, wheat gluten, casein etc., particularly the decomposition of soybean protein into amino acids has been carried out conventionally using acid hydrolysis with hydrochloric acid or sulfuric acid or the conventional proteases derived from microorganisms e.g. aspergillus (Japanese Patent LOP Publication No. 70,852/1976, Japanese Patent Publication No. 32,344/1980, Japanese Patent Publication No. 60,480/1991, Japanese Patent LOP Publication No. 239,966/1987, Japanese Patent LOP Publication No. 2392/1990 and Japanese Patent LOP Publication No. 112,461/1991).

The preparation of a soybean protein hydrolysate as a candidate for a natural seasoning by hydrolyzing soybean protein with an acid will require a reaction at 100 °C for 1 to 2 days, and such a high-temperature, long reaction needs high energy consumption. On the other hand, the acid hydrolysis of a protein is easy, but there are problems in that the resulting amino acids would also be decomposed (destroyed) and a high salt content would result from neutralization.

An approach to these problems was to decompose a protein under mild conditions by the conventional protease. However, the conventional protease, typically papain, subtilisin etc. are endopeptidases, and thus they decompose a protein into peptides but hardly decompose the peptides further into amino acids. Therefore, aspartic acid and glutamic acid participating strongly in tastes would hardly be released, while bitter tastes would be brought about, and in the present situation the resulting hydrolysate can thus not be used as a seasoning liquid.

To solve this problem, the combined use of exopeptidases i.e. a group of enzymes decomposing a peptide into amino acids, such as aminopeptidase, carboxypeptidase etc., is considered effective. For example, the importance of leucine aminopeptidase and acid carboxypeptidase is mentioned to increase a content of free amino acids for decomposing soybean protein with aspergillus, typically for making soy sauce by fermentation (Tadanobu Nakadai, Shouken, Vol. 11, No. 2, (1985)).

In this literature, however it is also described that in soy sauce there still remain dipeptide and tripeptide containing acidic amino acids in their sequences and these peptides are hardly decomposed with a peptidase derived from aspergillus. Further, these dipeptide and tripeptide also include a large number of peptides containing glutamic acid or aspartic acid at the N-terminal.

The "peptides hardly decomposed" herein used means that an enzyme serving as the catalyst for their decomposition, that is, a peptidase, has low substrate specificity for them.

This poor ability to decompose dipeptide and tripeptide containing acidic amino acids in their sequences is not only a problem with the peptidase derived from aspergillus in making soy sauce by fermentation, but also a problem with commercial peptidase preparations, typically those derived from aspergillus.

In the industry of soy sauce etc., therefore, there is a demand for the discovery of a certain peptidase effectively decomposing low-molecular-weight peptides containing glutamic acid and aspartic acid in their sequences in order to raise the degree of released amino acids in a peptide or protein hydrolysate.

The present inventors searched soybean cotyledons for a clue to solving the above problem. This is because from the fact that the proteins stored in soybeans are decomposed completely into amino acids in a very short time as the soybeans germinate, the present inventors expected the germinated soybeans to contain a certain peptidase which can easily decompose even hardly decomposable peptides derived from the storage proteins.

In particular, the present inventors speculated that there might be a certain peptidase with substrate specificity and properties by which glycinin and  $\beta$ -conglisinin, which are present as major storage proteins containing acidic amino acid-enriched motifs with e.g. successive acidic amino acids such as -Glu-Glu-Glu-Glu-Glu-, can be decomposed completely into amino acids.

As proteolytic enzymes found in germinated soybeans, the followings have already reported: 7S globulin protease (K. A. Wilson et al., Plant Physiol. 82, 71 (1986), X. Qi et al., Plant Physiol. 99, 725 (1992)), 11S globulin protease (K. A. Wilson et al., Plant Physiol. 82, 71 (1986), K. A. Wilson et al., Plant Physiol. 88, 355 (1988)), Bowman-Birk type trypsin inhibitor protease (M. A. Madden et al., Phytochemistry 24, 2811 (1985)), Qunitz type trypsin inhibitor protease (P. M. Hartl et al., Phytochemistry 25, 23 (1986), K. A. Wilson et al., Plant Physiol. 88, 355 (1988)), serine protease (M. Akhtaruzzaman et al., Biosci. Biotech. Biochem. 56(6), 878 (1992), novel thiol protease D3 (Japanese Patent LOP Publication No. 264/1996 published on January 8, 1996, and Japanese Patent Application No. 353,931/1995)).

However, these enzymes are endopeptidases and are thus not suitable for decomposing an acidic amino acid-containing peptide into amino acids.

Further, carboxypeptidase (Sachiho Kubota, Yakugaku Zasshi 96(5), 639 (1976)) and aminopeptidase (Shinji Watanabe et al., Nippon Nogei Kagakkaishi 63(3), 617 (1989)) have been reported as exopeptidase but these are not

suitable for decomposing an acidic amino acid-containing peptide into amino acids.

[Problem to Be Solved by the Invention]

5 Therefore, the object of the present invention is to provide an aminopeptidase effectively decomposing a low-molecular-weight peptide whose sequence contains glutamic acid and aspartic acid present in a soybean hydrolysate etc. as well as a method of hydrolyzing a peptide or protein by use of said aminopeptidase.

[Means to Solve the Problem]

10

As a result of their eager research to solve the above problem, the present inventor found that an enzyme which can be used for decomposing a peptide containing acidic amino acids is present in an extract from germinated soybean cotyledons, to complete the present invention.

15 The present invention provides aminopeptidase GX having the activity of easily decomposing a peptide or protein having acidic amino acids such as L- $\alpha$ -glutamylglutamic acid (Glu-Glu) etc. at the N-terminal.

The aminopeptidase GX of the present invention possesses the following properties:

- 1) optimum pH: about 5.5 to about 9.5;
- 2) optimum temperature: about 25 to about 60 °C;
- 20 3) temperature stability: keeping about 80 % or more activity after left at 50 °C for 80 minutes or about 40 % or more activity after left at 60 °C for 40 minutes;
- 4) molecular weight: about 400 to 550 kD (gel filtration), about 380 to 460 kD (native PAGE), and about 53 to 60 kD, about 30 to 32 kD,
- 25 and about 25 to 28 kD (SDS-PAGE after reduction and heating);
- 5) substrate specificity: decomposing a peptide or protein containing glutamic acid or aspartic acid at the N-terminal to release the glutamic acid or aspartic acid;
- 6) inhibitors: inhibited by leuhistin, actinonin, alphamenine A or 1,10-orthophenanthroline; and
- 7) effect of metal ions: inhibited by magnesium or copper.

30

A typical example of the aminopeptidase GX of the present invention is that derived from germinated soybean cotyledons.

In addition, the present invention provides a method of hydrolyzing a peptide or protein which comprises allowing said aminopeptidase GX to contact and react with a peptide or protein, as well as a method of hydrolyzing a peptide or protein which comprises allowing a cell extract obtained by disrupting germinated soybean cotyledons containing said aminopeptidase GX to contact and react with a peptide or protein.

35 Further, the present invention provides a method of hydrolyzing a peptide or protein which comprises allowing a combination of said aminopeptidase GX and protease D3 (Japanese Patent LOP Publication No. 264/1996 and Japanese Patent Application No. 353,931/1995) to contact and react with a peptide or protein.

40 Furthermore, the present invention provides a method of hydrolyzing a peptide or protein which comprises allowing a combination of aminopeptidase GX, protease D3 and leucine aminopeptidase to contact and react with a peptide or protein.

Said protease D3 is an enzyme purified from an extract of germinated soybean cotyledons and its properties and purification method were described in detail in Japanese Patent LOP Publication No. 264/1996. Recombinant protease D3 was also prepared in *E. coli* (Japanese Patent Application No. 353,931/1995), and its details will be described in Example 11.

The protease D3 used is preferably in the form of a high-purity preparation, but may be a crude enzyme prepared from germinated soybean cotyledons.

50 The leucine aminopeptidase used may be a commercially available enzyme (e.g. No. 1503 available from Sigma) but is preferably a group of crude daizdu (soybean) leucine aminopeptidases obtained from an extract of germinated soybeans. There are 2 kinds of aminopeptidases in an extract of germinated soybeans as the daizdu (soybean) leucine aminopeptidase (DLAP), that is DLAP1 and DLAP2. Said DLAP1 and DLAP2 are very similar in properties to those of soybean-derived aminopeptidase which has already been reported (Shinji Watanabe et al., Nippon Nogei Kagakkaishi 63(3), 617 (1989)), but there is no positive proof that they are the same. The preparation of DLAP1 and DLAP2 will be described in Example 14.

55 DLAP1 is an aminopeptidase with the following properties:

- 1) optimum pH: about 5.5 to about 9.5;
- 2) optimum temperature: about 25 to about 60 °C;

- 3) temperature stability: keeping about 90 % or more activity after left at 50 °C for 60 minutes, or keeping about 25 % or more activity after left at 60 °C for 10 minutes;
- 4) molecular weight: about 60 to 70 kD (native PAGE, activity staining);
- 5) substrate specificity: decomposing a peptide or protein containing leucine, alanine, glycine, phenylalanine, lysin, arginine, methionine, etc. at the N-terminal to release the corresponding amino acid; and
- 6) inhibitors: inhibited by amastatine or actinonin.

DLAP2 is an aminopeptidase with the following properties:

- 1) optimum pH: about 5.0 to about 9.0;
- 2) optimum temperature: about 30 to about 70 °C;
- 3) molecular weight: about 45 to 55 kD (native PAGE, activity staining);
- 5) substrate specificity: decomposing a peptide or protein containing phenylalanine etc. at the N-terminal to release the phenylalanine etc; and
- 6) inhibitors: inhibited by actinonin or phenylmethane sulfonyl fluoride.

[Best Embodiments for Carrying Out the Invention]

Hereinafter, the present invention is described in detail.

The type of germinated soybeans preferably used as a material for preparing the novel aminopeptidase of the present invention is not limited. That is, the cultivation area and variety of the soybeans are not limited; that is, commercially available soybeans, or those used as raw materials to extract oil, is not limited. There is no limitation to their germination method and cultivation conditions, and they can be used regardless of the presence or absence of sprouts and the term after their germination, but germinated soybeans preferably used as raw materials are those grown for 7 days after seeding were allowed to absorb water.

To prepare the novel aminopeptidase GX of the present invention, soybeans, preferably those described above, are used as a source to extract the enzyme from it. Preferably, cotyledons are removed from the soybeans and used as a source to extract the enzyme from it. For industrial application of the novel aminopeptidase of the present invention, this extract from soybeans can be used as such, or its crude or purified preparation can also be used.

The purification of the novel aminopeptidase of the present invention will be described in detail in Example 3. The preparation of a cell extract obtained by disrupting germinated soybean cotyledons will also be described in detail in the Examples below.

The novel aminopeptidase GX of the present invention will act on an protein or oligopeptide having glutamic acid or aspartic acid at the N-terminal, among usual proteins such as casein, bovine serum albumin, hemoglobin, soybean protein or peptides derived from the foregoing usual proteins, thus releasing said glutamic acid or aspartic acid. Further, the novel aminopeptidase GX will act on peptides generally hardly decomposed with exopeptidases (known aminopeptidase, carboxypeptidase), such as a dipeptide having an acidic amino acid at the N-terminal, a biologically active peptide having an acidic amino acid at the N-terminal, and  $\alpha$ -polyglutamic acid, to release their N-terminal acidic amino acid (see Examples 7, 8, 9 and 10 below).

The optimum pH of the novel aminopeptidase GX of the present invention is in the range of about 5.5 to 9.5, strictly about pH 7 to about 9 (see FIG. 1). Therefore, the peptide or protein shall be hydrolyzed in this pH range.

The optimum temperature of the novel aminopeptidase GX is in the range of about 25 to 60 °C, strictly about 35 to 55 °C (see FIG. 2). Therefore, the peptide or protein shall be hydrolyzed in this temperature range.

The temperature stability of the novel aminopeptidase GX maintains about 80 % or more of the original activity even after incubation at 50 °C for 80 minutes or about 40 % or more of the original activity after incubation at 60 °C for 40 minutes. Therefore, the hydrolysis reaction of the peptide or protein is carried out desirably at 50 °C or less in order to prevent the inactivation of the enzyme for a long time (see FIG. 3).

The effect of inhibitors on the novel aminopeptidase of the present invention was examined. The results indicated that the enzyme was strongly inhibited by aminopeptidase inhibitors such as leuhistine, actinonin etc. and weakly inhibited by alphamenine A, 1,10-orthophenanthroline (i.e. metal chelate compound), magnesium chloride, copper chloride etc. (see Table 1). As can be seen from this result, the novel aminopeptidase GX of the present invention is an aminopeptidase where a metal is involved in its activity expression.

The estimated molecular weight of the novel aminopeptidase of the present invention is about 390 to 400 kD as determined in SDS-PAGE when the sample was not reduced nor heated (see FIG. 4). On the other hand, when the sample was reduced and heated, the enzyme was separated in SDS-PAGE into subunits with molecular weights of about 53 to 60 kD, about 30 to 32 kD, and 25 to 28 kD respectively (see FIG. 5).

The analysis of molecular weight of the purified enzyme by a gel filtration column indicated that the activity peak was present in a molecular weight of about 400 to 550 kD (see FIG. 6). Further, the molecular weight estimated by native PAGE was in the range of about 380 to 460 kD (see FIG. 7).

To obtain (a) an amino acid and peptide mixture or (b) protein hydrolysate with a high content of acidic amino acids, the novel aminopeptidase GX of the present invention as such, or a cell extract obtained by disrupting germinated soybean cotyledons, may be allowed to act on peptides or a protein hydrolysate after treatment with various proteases. For example, the decomposition of L- $\alpha$ -glutamylglutamic acid (Glu-Glu) with an extract of germinated soybeans in 50 mM sodium phosphate buffer (pH 8.0) in the presence of 2 mM sodium azide at 37 °C will result in significant release of glutamic acid from the substrate Glu-Glu. This decomposition does also proceed efficiently even in deionized water.

To obtain an amino acid mixture with a high content of acidic amino acids released from soybean protein, not only the novel aminopeptidase GX of the present invention as such, but a cell extract obtained by disrupting germinated soybean cotyledons can also be used to decompose a peptide mixture, that is, a protein hydrolysate treated with the enzyme cysteine protease D3 from germinated soybeans.

Alternatively, the peptide mixture treated with the enzyme cysteine protease D3 from germinated soybeans may be hydrolyzed with leucine aminopeptidase in combination with the aminopeptidase GX of the present invention, to give a hydrolysate with amino acids in high yield. For example, if the protein hydrolysate treated with cysteine protease D3 is subjected to hydrolysis at 42 °C with a combination of the novel aminopeptidase GX with DLAP1 and DLAP2 (germinated soybean-derived leucine aminopeptidases) in amounts of 0.4 to 4 U GX, 10 to 130 U DLAP1 and 10 to 65 U DLAP2 per mg of the substrate in a solution adjusted to pH 8.0 with sodium hydroxide, the peptide will be decomposed to release various amino acids significantly (see the Examples)

The present specification also comprises Fig. 1 to 18 as enclosed. The following is a brief description of the drawings.

FIG. 1 shows a pH profile of aminopeptidase GX activity at different pH values.

FIG. 2 shows a profile of aminopeptidase GX activity at different temperatures.

FIG. 3 shows a profile of the stability of aminopeptidase GX activity at different temperatures.

FIG. 4 shows a profile in SDS-PAGE of the purified aminopeptidase GX stained with Coomassie, where the sample was not heated nor reduced.

FIG. 5 shows a profile in SDS-PAGE of the purified aminopeptidase GX stained with Coomassie, where the sample was not heated nor reduced.

FIG. 6 shows the analysis of the molecular weight of the purified aminopeptidase GX by gel filtration through Superose 6.

FIG. 7 shows a profile in SDS-PAGE of the purified aminopeptidase GX subjected to activity staining, where the sample was not heated nor reduced.

FIG. 8 shows a mass spectrum of a reaction solution in which chromogranin A was decomposed with aminopeptidase A, the structure of chromogranin A, and masses of its respective fragments.

FIG. 9 shows an increase in glutamic acid when  $\alpha$ -polyglutamic acid was decomposed with aminopeptidase GX.

FIG. 10 shows the (partial) construction of expression plasmid pGEMp $\beta$ .

FIG. 11 shows the (partial) construction of expression plasmid pGEMp $\beta$ .

FIG. 12 shows a profile in SDS-PAGE of the purified aminopeptidase DLAP1 subjected to activity staining, where the sample was not heated nor reduced.

FIG. 13 shows a profile in SDS-PAGE of the purified aminopeptidase DLAP2 subjected to activity staining, where the sample was not heated nor reduced.

FIG. 14 shows a pH profile of the activity of aminopeptidase DLAP1 activity at different pH values.

FIG. 15 shows a profile of aminopeptidase DLAP1 activity at different temperatures.

FIG. 16 shows a profile of the stability of aminopeptidase DLAP1 activity at different temperatures.

FIG. 17 shows a pH profile of aminopeptidase DLAP2 activity at different pH values.

FIG. 18 shows a profile of aminopeptidase DLAP2 activity at different temperatures.

#### [Examples]

Hereinafter, the present invention is illustrated by reference to the Examples.

#### (Example 1. Detection of Peptidase Activity)

The substrate used was dipeptide Glu-Glu, i.e. consisting of glutamic acids, and the glutamic acid released by the action of the enzyme was determined. The glutamic acid was quantitatively determined using conventional amino acid analysis means with ninhydrin or Yamasa Glutamic Acid Assay Kit commercially available. Unless otherwise specified, the following conditions and method were used.

0.05 ml enzyme solution, 0.02 ml of 50 mM L- $\alpha$ -glutamylglutamic acid, 0.02 ml of 100 mM N-2-hydroxyethylpiperazine-N'-2-ethanesulfonic acid (HEPES) buffer (pH 7.2), and 0.11 ml H<sub>2</sub>O were mixed and reacted at 37 °C for 20 minutes. The reaction was stopped by adding 0.05 ml of 50 % aqueous acetic acid, and the glutamic acid released in the reaction solution was quantitatively determined. The activity of the enzyme causing release of 2  $\mu$ M glutamic acid from L- $\alpha$ -glutamylglutamic acid per minute was assumed to be 1 Unit.

(Example 2. Search for L- $\alpha$ -Glutamylglutamic Acid (Glu-Glu) Decomposing Peptidase)

The above detection method was used to search for decomposition activity in an extract of germinated soybean cotyledons. First, the extract of cotyledons 7 days after germination was examined for Glu-Glu decomposition activity.

To cotyledons 7 days after germination, 4 g (8 cotyledons), was added a 5-fold excess volume (20 ml) of an ice-cold buffer (20 mM potassium phosphate buffer pH 7.0, 200 mM NaCl, 10 mM 2-mercaptoethanol, 2 mM  $\text{NaN}_3$ ), and the cotyledons were disrupted with a homogenizer, extracted, filtered through a gauze. The filtrate was centrifuged at 32,000 g for 30 minutes at 4 °C. The supernatant was filtered through a filter paper to give a crude extract. In this crude extract, the enzyme activity was present.

(Example 3. Purification of L- $\alpha$ -glutamylglutamic Acid (Glu-Glu) Decomposing Peptidase)

Because the L- $\alpha$ -glutamylglutamic acid (Glu-Glu) decomposition activity was present in the extract from cotyledons 7 days after germination, the enzyme with this activity was designated GX, and the purification of GX was carried out in the following manner.

## 1. Extraction

To 600 g cotyledons 7 days after seeding was added a 5-fold excess volume (3000 ml) of an ice-cold buffer (20 mM potassium phosphate buffer pH 7.0, 200 mM NaCl, 0.1 mM Pefabloc SC (4-(2-aminoethyl)-benzenesulfonyl fluoride HCl), 10  $\mu\text{M}$  E-64, 10 mM 2-mercaptoethanol, 2 mM  $\text{NaN}_3$ ), and the cotyledons were disrupted with a homogenizer, extracted, and filtered through a gauze. The filtrate was centrifuged at 32,000 g for 30 minutes at 4 °C. The supernatant was filtered through a filter paper (No. 514 A, available from Advantech Toyo K.K.). The crude extract obtained as the filtrate was concentrated in a membrane concentration unit (Minitan, available from Millipore).

## 2. Ammonium Sulfate Fractionation

The pH of the crude extract thus prepared was confirmed (adjusted to pH 7.0 with NaOH), and this solution was made 40 % saturation with ammonium sulfate by adding ammonium sulfate (763 g/3150 ml). The solution was stirred at 4 °C for 6 hours and centrifuged at 32,000 g for 30 minutes at 4 °C to give a supernatant. The supernatant was made 65 % saturation with ammonium sulfate by further adding ammonium sulfate (564 g/3,400 ml). It was stirred at 4 °C overnight (15 hours) and then centrifuged at 32,000 g for 30 minutes at 4 °C to recover precipitates. The precipitates were dissolved in 120 ml buffer (50 mM potassium phosphate buffer pH 7.0, 100 mM NaCl, 2 mM  $\text{NaN}_3$ ) and dialyzed repeatedly against a buffer (2L x 5, 16 hours) at 7 °C. After dialysis, the dialysate was centrifuged at 32,000 g for 20 minutes at 4 °C and then filtered thorough a 0.2  $\mu\text{m}$  filter (Nalgene filterware) to be applied to anion exchange chromatography in the subsequent step.

## 3. Anion Exchange Chromatography

The sample obtained in 2 above was fractionated through HiLoad 26/10 Q Sepharose HP in FPLC system (Pharmacia) as follows. About 190 ml sample was divided due to a large volume into 8 aliquots and each aliquot was subjected to this ion-exchange chromatography. The column was previously equilibrated with the same buffer as used in the previous dialysis, and after application of the sample, the column was washed with a 5-fold column volume of the buffer. After washing, the adsorbed active fraction was eluted with an increasing linear gradient of from 100 mM to 250 mM NaCl in an 8-fold column volume of the buffer. The active fraction was collected (2000 ml) and concentrated to 90 ml in an ultrafiltration unit (Minitan, manufactured by Millipore). The sample was adjusted to 1 M ammonium sulfate by adding 11.88 g ammonium sulfate. The solution was centrifuged at 32,000 g for 10 minutes at 4 °C to give a supernatant to be applied to hydrophobic chromatography in the subsequent step.

## 4. Hydrophobic Chromatography

The sample obtained in 3 above was fractionated thorough HiLoad 26/10 Phenyl Sepharose HP in the FPLC system as follows. The sample was divided into 2 aliquots and each aliquot was applied to the column. The column was previously equilibrated with a buffer (50 mM potassium phosphate buffer pH 7.0, 1 M  $(\text{NH}_4)_2\text{SO}_4$ , 2 mM  $\text{NaN}_3$ ), and after addition of the sample, the column was washed with a 5-fold column volume of the buffer. Then, the adsorbed active component was fractionated in a decreasing linear gradient of from 1 M to 0 M ammonium sulfate in the buffer. The active fraction, 180 ml, was concentrated to 9.8 ml by an ultracentrifugation unit (Minitan (Millipore) and Centriprep 10 (Amicon)). The condensate was centrifuged at 32,000 g for 10 minutes at 4 °C to give a supernatant to be applied to gel filtration in the subsequent step.

## 5. Gel Filtration Chromatography

The sample obtained in 4 above was fractionated by gel filtration through HiLoad 26/60 Superdex 200pg in the FPLC system as follows. The sample was divided into 2 aliquots and each aliquot was subjected to chromatography. The column was previously equilibrated with a buffer (50 mM potassium phosphate buffer pH 7.0, 100 mM NaCl, 2 mM NaN<sub>3</sub>) and the sample was fractionated through it. The active fraction was concentrated to give a purified enzyme. The molecular weight of the purified enzyme was estimated to be 390 to 400 kD in SDS-PAGE when not subjected to reduction nor thermal treatment (see FIG. 4). When subjected to reduction and thermal treatment, the enzyme was separated into subunits having molecular weights of about 53 to 60 kD, about 30 to 32 kD and about 25 to 28 kD in SDS-PAGE respectively (see FIG. 5). When the purified enzyme was analyzed for molecular weight further by gel filtration on a gel filtration column Superose 6 (Pharmacia), an active peak appeared in a molecular weight of about 400 to 550 kD (see FIG. 6).

An increase in specific activity by a series of the purification steps was determined. The specific activity of the above crude GX enzyme solution for decomposition of Glu-Glu was about 7.6 mU/mg, while the specific activity of the purified enzyme GX was about 66 U/mg. This result indicated that in a series of the purification steps, the aminopeptidase GX was purified as highly as about 8700-fold in terms of specific activity.

Activity staining of the enzyme in native PAGE indicated that the molecular weight of the active enzyme was about 380-460 kD. Hereinafter, the active staining method is described.

## 20 (Example 4. Peptidase Activity Staining in Native PAGE)

20, 60, 120 mU enzyme was prepared in a Davis' sample preparation buffer (0.0625 M Tris-HCl buffer pH 6.8, 15 % glycerol, 0.001 % bromophenol blue) and electrophoresed in a commercial gel (Multigel 2/15, Dai-ichi Kagaku Yakuhin K.K.) by using an electrophoresis buffer (3 g Tris, 14.4 g/l glycine) at 5 °C. The gel after electrophoresis was mixed with solution A (800 µl of 1 M HEPES, pH 7.0, 300 µl of L-amino acid oxidase (1 mg/ml), 12.5 U peroxidase, 4 µl of 3-amino-4-ethylcarbazole (20 mM), 530 µl of 4-aminoantipyrine, 5840 µl of Glu-Phe (20 mM)) and solution B (8 ml of 2 % aqueous agarose solution (55 °C)) and incubated in an incubator at 37 °C for 2 hours. The L-amino acid oxidase and peroxidase were purchased from Boehringer Mannheim, the 3-amino-4-ethylcarbazole were purchased from Dojin Kagaku K.K. and all the other reagents were purchased from Nakarai Tesque K.K. The results indicated that the active enzyme was found at a molecular weight of about 380 to 460 kD (see FIG. 7).

The enzymatic properties of the purified GX thus obtained are described below.

## (Example 5. Determination of Optimum pH and Optimum Temperature of Aminopeptidase GX)

35 A change in enzyme activity (optimum pH) due to reaction pH was determined in the following manner.

The enzyme reaction buffers used were sodium acetate buffer (pH 4.0, 4.5, 5.0, 5.5, 6.0), potassium phosphate buffer (pH 6.0, 7.0), Tris-HCl buffer (pH 7.0, 7.8, 8.3, 8.8), and sodium carbonate buffer (pH 9.0, 9.5, 10.0).

180 µl of 3.5 mU GX was prepared in 50 mM buffer at each pH and pre-incubated for 5 minutes at 30 °C. 20 µl of the substrate Glu-Glu (5 mM for reaction) was added to each sample. It was stirred and incubated for 20 minutes. 20 µl of 1 M sodium acetate buffer (pH 4.0) was added to stop the reaction.

The measurement results are shown in FIG. 1. As can be seen from FIG. 1, the optimum pH of the aminopeptidase GX of the present invention is in the range of about 5.5 to 9.5, strictly about 7 to about 9 (see FIG. 1).

The measurement of a change in enzyme activity due to reaction temperature (optimum temperature) was carried out as follows:

45 20 µl of the substrate Glu-Glu (5 mM for reaction) was added to 50 mM sodium acetate buffer (pH 6.0) and 180 µl of thus obtained reaction solution was pre-incubated for 5 minutes at each temperature (25, 30, 37, 42, 50, 60, 70 °C). 7 mU aminopeptidase GX was further added to it. This sample was stirred and incubated for 20 minutes. 20 µl of 1 M sodium acetate buffer (pH 4.0) was added to stop the reaction.

50 The measurement results are shown in FIG. 2. As can be seen from FIG. 2, the optimum temperature of the aminopeptidase GX of the present invention is in the range of about 25 to 60 °C, strictly about 35 to 55 °C (see FIG. 2).

The determination of temperature stability was carried out as follows:

55 3.5 U/ml enzyme was incubated in a buffer (pH 6.0) at predetermined temperatures (25, 30, 37, 42, 50, 60, 70 °C) for predetermined times (10, 20, 30, 40, 80 minutes) and then examined for its remaining activity at 30 °C.

The measurement results are shown in FIG. 3. The enzyme of the present invention maintains about 80 % or more activity even after 80-minute incubation at 50 °C or about 40 % or more activity after 40-minute incubation at 60 °C.

Therefore, the reaction temperature for a long reaction is preferably 50 °C or less.

(Example 6. Effect of Protease Inhibitors on GX)

The effect of inhibitors and metals on the novel aminopeptidase GX of the present invention was examined. The inhibitors used were actinonin, amastatin, antipain, alphamenine A, diprotin A, leuhistin, phenylmethane sulfonyl fluoride (PMSF), trans-epoxysuccinyl-L-leucylamide (4-guanidino)-butane (E-64), iodoacetamide, and 1,10-orthophenanthroline, and the metals used were zinc chloride, manganese chloride, magnesium chloride and copper chloride. The 1,10-orthophenanthroline, zinc chloride, manganese chloride, magnesium chloride and copper chloride were purchased from Nakarai Tesque K.K. and all the other inhibitors were purchased from Peptide Institute, Inc.

In the presence of each inhibitor, the enzyme was left at room temperature (25 °C) for 20 minutes, and the remaining activity of the enzyme was determined in the same manner as in Example 1.

The results of activity measurement in the presence of each inhibitor are shown in Table 1. The aminopeptidase GX of the present invention was strongly inhibited by aminopeptidase inhibitors such as leuhistin and actinonin and weakly inhibited by alphamenine A, 1,10-orthophenanthroline (i.e. metal chelate compound), magnesium chloride and copper chloride.

As can be seen from the foregoing, the aminopeptidase GX of the present invention is an aminopeptidase where a metal is involved in its activity expression.

Table 1

Effect of Each Protease Inhibitor on GX		
Inhibitor	Concentration	Remaining Activity (Relative Value %)
-	-	100
actinonin	100 $\mu$ M	12
amastatin	100 $\mu$ M	100
antipain	100 $\mu$ M	100
alphamenine A	100 $\mu$ M	63
diprotin A	100 $\mu$ M	100
E-64	10 $\mu$ M	100
leuhistin	100 $\mu$ M	0
PMSF	10 $\mu$ M	100
iodoacetamide	50 mM	100
orthophenanthroline	10 $\mu$ M	47
zinc chloride	2 mM	76
manganese chloride	2 mM	88
magnesium chloride	2 mM	32
copper chloride	2 mM	56

(Example 7. Substrate Specificity of GX for Dipeptide)

To examine the substrate specificity of GX, the activity of decomposing each dipeptide was compared with the activity of decomposing Glu-Glu. The dipeptides used are Glu-Asp, Glu-Ser, Glu-Thr, Glu-Phe, Glu-Ala, Glu-Gly, Glu-Lys,  $\gamma$ -Glu-Glu,  $\gamma$ -Glu-Leu, Ser-Glu, Ala-Glu, Phe-Glu, Lys-Glu, Pro-Glu, Asp-Asp, Asp-Phe, Asp-Glu, Asp-Ala, Asp-Lys, Asp- $\epsilon$ -Lys, Gln-Gly and Gln-Gln, and the enzyme was allowed to act on each dipeptide in the same manner as in Example 1 using Glu-Glu as the substrate, then the amino acids in the reaction solution were quantified in amino acid analysis to compare the decomposition activity.

The results of the relative decomposition activity of GX for each dipeptide are shown in Table 2.



Table 2

Relative Decomposition Activity of GX for Each Dipeptide	
Dipeptide	Relative Decomposition Activity (%)
Glu-Glu	100
Glu-Lys	175
Glu-Gly	80
Glu-Ala	108
Glu-Phe	32
Glu-Thr	102
Glu-Ser	128
Glu-Asp	63
$\gamma$ -Glu-Glu	4
Pro-Glu	0
Lys-Glu	0
Phe-Glu	0
Ala-Glu	0
Ser-Glu	0
Asp- $\epsilon$ -Lys	11
Asp-Lys	172
Asp-Ala	110
Asp-Glu	30
Asp-Phe	79
Asp-Asp	14
Gln-Gln	0
Gln-Gly	0
$\gamma$ -Glu-Leu	0

(Example 8. Substrate Specificity of GX for Amino Acid Paranitroanilide Derivative, and Comparison with Aminopeptidase M)

To examine the substrate specificity of GX, the substrate specificity for each amino acid paranitroanilide derivative was compared with that for glutamic acid paranitroanilide (Glu-pNA). 0.05 ml enzyme solution, 0.02 ml of 2 mM each amino acid paranitroanilide, 0.02 ml of 100 mM HEPES buffer (pH 7.0), and 0.11 ml H<sub>2</sub>O were mixed and reacted at 37 °C for 20 minutes. The reaction was stopped by adding 0.05 ml of 50 % aqueous acetic acid, and the amount of paranitroaniline released in the reaction solution was determined. For comparison, the enzyme activity causing release of 1  $\mu$ mole paranitroaniline per minute from each amino acid paranitroanilide derivative was assumed to be 1 Unit. Its substrate activity was compared with that of aminopeptidase M (APase M) which is an enzyme commercially available from Pierce company.

The results of the decomposition activity for amino acid paranitroanilide derivatives are shown in Table 3.

Table 3

Synthetic Substrate	GX U/mg	APase M (U/mg)
Glu-pNA	65	2.8
Leu-pNA	0	1500
Phe-pNA	0	1100
Gly-pNA	0	100
Ala-pNA	0	2300
Pro-pNA	0	0.3

(Example 9. Decomposition of Chromogranin A by GX)

The decomposition of chromogranin A (Peptide Institute Inc.) i.e. an oligopeptide containing glutamic acid at the N-terminal by GX was examined.

0.02 ml buffer (500 mM ammonium carbonate), 0.05 ml of 5 mM chromogranin A, 0.08 ml H<sub>2</sub>O, and 0.05 ml of 6 mU enzyme were reacted at 37 °C. The reaction solution was sampled with time in an amount of 0.02 ml for each sampling. The reaction was stopped with 0.02 ml of 50 % aqueous acetic acid. The reaction solution was analyzed for molecular weight by laser mass spectrum (Kompakt Maldi III, manufactured by Shimadzu Kratos).

FIG. 8 shows a profile of the reaction solution after 15-hour reaction. The molecular weights of 1823.1, 1693.7, 1564.0, 1434.8, 1305.7, and 1176.1 was observed and these results corresponded to the molecular weights of chromogranin A and its fragments. GX had the aminopeptidase activity by which glutamic acid was released one by one from chromogranin A.

(Example 10. Decomposition of  $\alpha$ -polyglutamic Acid by GX)

The decomposition of  $\alpha$ -polyglutamic acid (molecular weight of 8000 or more) (Peptide Kenkyusho K.K.) by GX was examined.

0.05 ml of 100 mM HEPES buffer, 0.05 ml  $\alpha$ -polyglutamic acid (50 mg/ml), 0.3 ml H<sub>2</sub>O, and 0.1 ml of 100 mU enzyme were reacted at 37 °C. The reaction solution was sampled with time in an amount of 0.02 ml for each sampling. The reaction was stopped with 0.02 ml of 50 % aqueous acetic acid. The content of glutamic acid in the reaction solution was determined. The results are shown in FIG. 9.

Hereinafter, a preparation method for protease D3 derived from germinated soybean cotyledons, which was used in combination for decomposition of soybean protein, is described.

The method of detecting protease D3 derived from germinated soybean cotyledon, the definition of its activity unit, and the preparation method for this enzyme from an extract of germinated soybean cotyledons were described in detail in pending Japanese Patent LOP Publication No. 264/1996 filed by the present inventors.

Protease D3 derived from germinated soybean cotyledons is a novel thiol protease which is enzyme 1 or 2 described in Japanese Patent LOP Publication No. 264/1996 and either of the two can be used to decompose soybean protein.

In connection with protease D3, Japanese Patent Application No. 353,931/1995 filed on December 28, 1995 by the present inventors, describes a process for producing recombinant protease D3 by use of cDNA and *E. coli* containing said cDNA. Hereinafter, the recombinant protease D3 by use of *E. coli* is described.

As the recombinant protease D3 produced by *E. coli*, proteases D3- $\alpha$  and D3- $\beta$  are described in Japanese Patent Application No. 353931/1995 and either of the two can be used to decompose soybean protein.

(Example 11. Preparation of Recombinant Protease D3 by Use of *E. coli*)

A cDNA library was prepared from mRNA derived from germinated soybean cotyledons, and a part of protease D3- $\beta$  cDNA from this library was cloned in a usual manner and integrated into an expression vector capable of functioning in *E. coli*. *E. coli* transformed with this expression vector was cultured to yield a gene product as a protein inclusion body. This inclusion body was removed from the microorganism, then lyzed and unwound *in vitro* and allowed to be in contact with a substrate with the pH shifted to an acid side. Hereinafter, this method is described in detail.

The construction of the expression plasmid is shown in FIGS. 10 and 11.

As shown in FIG. 10, clone pMOSS-a into which the whole-length D3- $\beta$  cDNA had been integrated was cleaved with restriction enzyme SphI, then blunt-ended with Klenow enzyme, and next cleaved with EcoRI to give an about 870 bp fragment. A commercial expression vector pGEMEX-1 (Promega) was cleaved with NheI, then blunt-ended with Klenow enzyme, and cleaved with EcoRI to give a large fragment. The above 870 kb fragment was ligated to this large fragment to give plasmid pGEMp $\beta$ -N.

Separately, as shown in FIG. 11, pMOSS-a was cleaved with EcoRI and PvuII to give a 190 bp fragment which was then inserted between EcoRI and SmaI sites on M13mp19 to construct M13 $\beta$ -C. Using a single-stranded DNA of M13 $\beta$ -C as a template and the oligonucleotide 5'-CAAATGTCTGAGACAACACT-3' as a mutagenesis primer, TGT (Cys codon at the 242-position in an open reading frame of D3- $\beta$  cDNA) was mutated by site-directed mutagenesis into TGA (termination codon) in a sculptor *in vitro* mutagenesis system (Amersham company) to give M13 $\beta$ -C (C242Opal). It was then cleaved with EcoRI and BamHI to give an about 190 bp fragment.

Thereafter, as shown in FIG. 10, a large fragment obtained by cleaving said plasmid pGEMp $\beta$ -N with EcoRI and BamHI was ligated to the about 190 bp fragment obtained from above M13 $\beta$ -C (C242Opal), whereby an expression vector pGEMp $\beta$  was constructed. That is, the sequence integrated into this pGEMp $\beta$  so as to be expressed is as shown in SEQ ID NO: 1 in the Sequence Listing, where nucleotides 1 to 9 were derived from the vector.

Then, said pGEMp $\beta$  was transformed into *E. coli* JM109 (DE3) (Promega). The resulting transformant was cultured at 37 °C under shaking in a suitable medium containing isopropyl- $\beta$ -D-thiogalactopyranoside until a D3 gene product was accumulated as a protein inclusion body in the microorganism.

The microorganism thus cultured was collected, disrupted by ultrasonication and centrifuged to recover the protein inclusion body. This protein inclusion body was washed and dissolved in 50 mM Tris-HCl buffer, pH 8 containing 8 M urea, 10 mM dithiothreitol, 50 mM NaCl and 5 mM ethylenediaminetetraacetic acid (EDTA) to give an about 10 mg/ml protein solution. This solution was named solubilized pD3- $\beta$ .

To 1 part of the solubilized pD3- $\beta$  were slowly added 100 parts of a solution containing reduced glutathione and oxidized glutathione (i.e. 50 mM potassium phosphate (pH 10.5) containing 1 mM reduced glutathione, 0.1 to 0.5 mM oxidized glutathione and 5 mM EDTA) but not containing any protein modifier.

The mixture was concentrated about 50-fold and the solvent was replaced by 5 mM potassium phosphate buffer (pH 10) containing 200 mM NaCl. This solution was named refolded pD3- $\beta$ .

The specific activity of the resulting D3- $\beta$  for decomposition of c30 was 15 U/mg. The measurement of this activity was described in detail in Japanese Patent LOP Publication No. 264/1996.

pGEMp $\beta$  was transformed into *E. coli* JM109, and this transformant has been deposited since July 3, 1995 with the National Institute of Bioscience and Human Technology, Agency of Industrial Science and Technology, Japan (accession number: FERM BP-5793).

Hereinafter, an activity detection method, preparation method etc. are described for leucine aminopeptidase derived from germinated soybean cotyledons, which was used in combination to decompose soybean protein.

#### (Example 12. Activity Detection for Leucine Aminopeptidase)

The substrate used was leucine paranitroanilide, and the paranitroanilide released by the action of this enzyme was determined. Unless otherwise specified, the following conditions and method were used.

0.05 ml enzyme solution, 0.02 ml of 2 mM L-leucine paranitroanilide, 0.02 ml of 100 mM HEPES buffer (pH 7.0) and 0.11 ml H<sub>2</sub>O were mixed and reacted at 37 °C for 20 minutes. The reaction was stopped by adding 0.05 ml of 50 % aqueous acetic acid, and the amount of paranitroaniline released in the reaction solution was determined. The enzyme activity causing release of 1  $\mu$ mole paranitroaniline per minute from leucine paranitroanilide as the substrate was assumed to be 1 Unit.

#### (Example 13. Search for Leucine Aminopeptidase)

The above detection method was used to search for decomposition activity in an extract of germinated soybean cotyledons. First, cotyledons 7 days after germination were examined for leucine paranitroanilide decomposition activity.

To cotyledons 7 days after seeding, 4 g (8 cotyledons), was added a 5-fold excess volume (20 ml) of an ice-cold buffer (20 mM potassium phosphate buffer pH 7.0, 200 mM NaCl, 10 mM 2-mercaptoethanol, 2 mM NaN<sub>3</sub>), and the cotyledons were disrupted with a homogenizer, extracted, and filtered through a gauze. The filtrate was filtered at 32,000 g for 30 minutes at 4 °C. The supernatant was filtered through a filter paper to give a crude extract. In this crude extract, the enzyme activity was present.

(Example 14. Purification of Daidzu (Soybean) Leucine Aminopeptidase)

Because the leucine paranitroanilide (Leu-pNA) decomposition activity was present in the extract from cotyledons 7 days after germination, the enzyme with this activity was designated daidzu leucine aminopeptidase (DLAP), and the purification of DLAP fractions (DLAP1 and DLAP2) was carried out in the following manner.

#### 1. Extraction

To 600 g cotyledons 7 days after seeding was added a 5-fold excess volume (3000 ml) of an ice-cold buffer (20 mM potassium phosphate buffer pH 7.0, 200 mM NaCl, 0.1 mM Pefabloc SC (4-(2-aminoethyl)-benzenesulfonyl fluoride HCl), 10  $\mu$ M E-64, 10 mM 2-mercaptoethanol, 2 mM  $\text{NaN}_3$ ), and the cotyledons were disrupted with a homogenizer, extracted and filtered through a gauze. The filtrate was centrifuged at 32,000 g for 30 minutes at 4 °C. The supernatant was filtered through a filter paper (No. 514 A, available from Advantec Toyo K.K.). The crude extract obtained as the filtrate was concentrated in a membrane concentration unit (Minitan, available from Millipore company).

#### 2. Ammonium Sulfate Fractionation

The pH of the crude extract thus prepared was confirmed (adjusted to pH 7.0 with NaOH), and this solution was made 40 % saturation with ammonium sulfate by adding ammonium sulfate (763 g/3150 ml). The solution was stirred at 4 °C for 6 hours and centrifuged at 32,000 g for 30 minutes at 4 °C to give a supernatant. The supernatant was made 65 % saturation with ammonium sulfate by further adding ammonium sulfate (564 g/3,400 ml). It was stirred at 4 °C overnight (15 hours) and then centrifuged at 32,000 g for 30 minutes at 4 °C to recover precipitates. The precipitates were dissolved in 120 ml buffer (50 mM potassium phosphate buffer pH 7.0, 100 mM NaCl, 2 mM  $\text{NaN}_3$ ) and dialyzed repeatedly against a buffer (2L x 5, 16 hours) at 7 °C. After dialysis, the dialysate was centrifuged at 32,000 g for 20 minutes at 4 °C and then filtered through a 0.2  $\mu$ m filter (Nalgene filterware) to be applied to anion exchange chromatography in the subsequent step.

#### 3. Anion Exchange Chromatography

The sample obtained in 2 above was fractionated through HiLoad 26/10 Q Sepharose HP in the FPLC system (Pharmacia) as follows. About 190 ml sample was divided due to a large volume into 8 aliquots and each aliquot was subjected to this ion-exchange chromatography. The column was previously equilibrated with the same buffer as used in the previous dialysis, and after application of the sample, the column was washed with a 5-fold column volume of the buffer. Just after application of the sample, the eluted fraction including the wash, about 2,000 ml in total, was concentrated to 55 ml in an ultracentrifugation unit (Minitan, manufactured by Millipore). This fraction was designated DLAP1 fraction. This sample was adjusted to 1 M ammonium sulfate by adding 7.26 g ammonium sulfate. The solution was centrifuged at 32,000 g for 20 minutes at 4 °C to give a supernatant to be applied to hydrophobic chromatography in the subsequent step. After washing, the active fraction remaining on the column was eluted with a linear gradient of from 100 mM to 300 mM NaCl in a 8-fold column volume of the buffer. The active fraction adsorbed in the column in this step was designated DLAP2 fraction. The DLAP2 fraction was eluted with 200 to 300 mM NaCl in the buffer, and the eluent, about 410 ml in total, was concentrated to 46 ml in an ultrafiltration unit (Minitan, Millipore). This sample was adjusted to 1 M ammonium sulfate by adding 6.07 g ammonium sulfate. This solution was centrifuged at 32,000 g for 20 minutes at 4 °C to be applied to hydrophobic chromatography in the subsequent step.

#### 4. Hydrophobic Chromatography

The samples DLAP1 and DLAP2 obtained in 3 above were further fractionated with HiLoad 26/10 Phenyl Sepharose HP in the FPLC system as follows. DLAP1, about 60 ml, was divided into 2 aliquots and each aliquot was applied to the column. The column was previously equilibrated with a buffer (50 mM potassium phosphate buffer pH 7.0, 1 M  $(\text{NH}_4)_2\text{SO}_4$ , 2 mM  $\text{NaN}_3$ ), and after application of the sample, the column was washed with a 5-fold column volume of the same buffer. Then, the active component was fractionated by eluting it in a decreasing linear gradient of from 1 M to 0 M ammonium sulfate in the buffer. In this step, the active fraction was eluted with 0.4 to 0 M sulfate ammonium. This eluate, 600 ml, was concentrated to 3.0 ml with Minitan and Centriprep 10 (Amicon). The condensate was centrifuged at 32,000 g for 10 minutes at 4 °C to give a supernatant to be subjected to gel filtration in the subsequent step. DLAP2, about 46 ml, was applied to the column. The column was previously equilibrated with a buffer (50 mM potassium phosphate buffer pH 7.0, 1 M  $(\text{NH}_4)_2\text{SO}_4$ , 2 mM  $\text{NaN}_3$ ), and after application of the sample, the column was washed with a 5-fold column volume of the same buffer. Then, the active component was fractionated by eluting it in a decreasing linear gradient of from 1 M to 0 M ammonium sulfate in the buffer. In this step, the active fraction was eluted with 0.4 M to 0 M sulfate ammonium. This eluate, 100 ml, was concentrated to 1.0 ml with Minitan and Centriprep 10

(Amicon). The condensate was centrifuged at 32,000 g for 10 minutes at 4 °C to give a supernatant to be subjected to gel filtration in the subsequent step.

## 5. Gel filtration chromatography

The DLAP1 obtained in above 4 was fractionated by gel filtration in HiLoad 26/60 Superdex 200pg in the FPLC system as follows. The sample was divided into 2 aliquots and each aliquot was applied to chromatography. The column was equilibrated with a buffer (50 mM potassium phosphate buffer pH 7.0, 100 mM NaCl, 2 mM NaN<sub>3</sub>) prior to fractionation of the sample. The obtained active fraction was concentrated to give a DLAP1 preparation.

The DLAP2 obtained in 4 above was fractionated by gel filtration through the same column in the FPLC. The obtained active fraction was concentrated to give a DLAP2 preparation.

An increase in specific activity by a series of the purification steps was determined. The specific activity of the above crude leucine aminopeptidase enzyme was about 5 U/mg, while the specific activity of the purified enzyme DLAP1 was about 160 U/mg, indicating that DLAP1 was purified about 32-fold in terms of specific activity in the purification process. Because the specific activity of the purified DLAP2 was also about 160 U/mg, DLAP2 was purified 32-fold in terms of specific activity. DLAP1 was detected more sensitively with alanine paranitroanilide (Ala-pNA) than Leu-pNA; that is, its specific activity was 270 U/mg in terms of activity unit using alanine paranitroanilide (Ala-pNA).

Hereinafter, the activity of DLAP1 will be expressed in activity measurements using alanine paranitroanilide. DLAP2 was detected more sensitively by use of phenylalanine paranitroanilide (Phe-pNA) than Leu-pNA; that is, its specific activity was 3200 U/mg in terms of activity unit using phenylalanine paranitroanilide (Phe-pNA).

Hereinafter, the activity of DLAP2 will be expressed in activity measurements using phenylalanine paranitroanilide. From the activity staining of the enzyme in native PAGE, the molecular weight of active DLAP1 enzyme was estimated to be about 60 to 70 kD (see FIG. 12). From similar staining, the molecular weight of active DLAP2 enzyme was estimated to be about 45 to 55 kD (see FIG. 13). Hereinafter, the activity staining method is described.

### (Example 15. Activity Staining of DLAP1 and DLAP2 in Native PAGE)

20, 60 and 120 mU enzyme was prepared in a Davis' sample preparation buffer (0.0625 M Tris-HCl buffer pH 6.8, 15 % glycerol, 0.001 % bromophenol blue) and electrophoresed in a commercial gel (Multigel 2/15, Dai-ichi Kagaku Yakuhin K.K.) in an electrophoresis buffer (3 g Tris, 14.4 g/l glycine) at 5 °C. To detect DLAP1, the gel after electrophoresis was incubated in solution A (50 mM potassium phosphate buffer pH 8.0, 0.5 mg/ml alanine-β-naphthylamide) in an incubator at 37 °C for 30 minutes. After washing with distilled water, the gel was stained with solution B (50 mM potassium phosphate buffer pH 8.0, 1 mg/l Fast Blue B Salt (o-dianisidine, tetrazotized). Fast Blue B Salt was supplied from Sigma, and all the other reagents from Nakarai Tesque K.K. From the result, the molecular weight of active DLAP1 enzyme was estimated to be about 60 to 70 kD (see FIG. 12).

DLAP2 was similarly detected using 50 mM potassium buffer pH 8.0 and 0.5 mg/ml phenylalanine-β-naphthylamide as solution A. From the result, the molecular weight of active DLAP2 enzyme was estimated to be about 45 to 55 kD (see FIG. 13).

The enzymatic properties of DLAP1 obtained in this manner are described below. The activity of DLAP1 was determined using L-alanine paranitroanilide as the substrate in place of L-leucine paranitroaniline used in the method of Example 12.

### (Example 16. Determination of Optimum pH and Optimum Temperature of DLAP1)

A change in enzyme activity due to reaction pH (optimum pH) was determined in the following manner.

The enzyme reaction buffers used were sodium citrate buffer (pH 3.0, 3.5, 4.0, 4.5, 5.0, 5.5, 6.0), MES buffer (pH 5.5, 6.0, 6.5, 7.0), HEPES buffer (pH 7.0, 7.5, 8.0), Tris-HCl buffer (pH 8.0, 8.5, 9.0), and sodium carbonate buffer (pH 9.0, 9.5, 10.0, 10.5).

180 μl of 0.7 U DLAP1 was prepared in 50 mM buffer at each pH and pre-incubated for 5 minutes at 30 °C. 20 μl of the substrate L-alanine paranitroanilide (0.4 mM for reaction) was added to each sample, then stirred and incubated for 20 minutes. 50 μl of 1 M sodium acetate buffer (pH 4.0) was added to stop the reaction.

The measurement results are shown in FIG. 14. As can be seen from this graph, the optimum pH of DLAP1 of the present invention is in the range of about 5.5 to 9.5, strictly about 7 to about 9 (see FIG. 14).

A change in enzyme activity due to reaction temperature (optimum temperature) was determined as follows:

20 μl of the substrate L-alanine paranitroanilide (0.4 mM for reaction) was added to 50 mM MES buffer (pH 6.0) and 180 μl of thus obtained reaction solution was pre-incubated for 5 minutes at each temperature (25, 30, 37, 42, 50, 60, 70 °C). 0.7 U DLAP1 was further added to it.

This sample was stirred and then incubated for 20 minutes. 50  $\mu$ l of 1 M sodium acetate buffer (pH 4.0) was added to it to stop the reaction.

The measurement results are shown in FIG. 15. As can be seen from this graph, the optimum temperature of the DLAP1 of the present invention is in the range of about 25 to 60 °C, strictly about 35 to 55 °C (see FIG. 15).

The determination of temperature stability was carried out as follows:

900 U/ml enzyme was incubated in a buffer (pH 6.5) at predetermined temperatures (30, 42, 50, 60 °C) for predetermined times (10, 20, 40, 60 minutes) and then examined for its remaining activity at 30 °C.

The measurement results are shown in FIG. 16. The DLAP1 of the present invention maintained about 90 % or more of the original activity even after 60-minute incubation at 50 °C and the activity was maintained about 25 % or more after 10-minute incubation at 60 °C. Therefore, the reaction temperature for longer reaction is preferably 50°C or less.

#### (Example 17. Effect of Protease Inhibitors on DLAP1)

The effect of protease inhibitors on the DLAP1 of the present invention was examined. The inhibitors used were actinonin, amastatin, antipain, alphamenine A, diprotin A, leuhistin, phenylmethane sulfonyl fluoride (PMSF), transepoxy succinyl-L-leucylamide (4-guanidino)-butane (E-64), iodoacetamide, 1,10-orthophenanthroline, and N-ethylmaleimide (NEM). The 1,10-orthophenanthroline and N-ethylmaleimide were purchased from Nakarai Tesque K.K. and all the other inhibitors were purchased from Peptide Kenkyusho K.K.

In the presence of each inhibitor, the enzyme was left at room temperature (25 °C) for 20 minutes, and the remaining activity of the enzyme was determined in the same manner as in Example 12.

The results of activity measurements in the presence of each inhibitor are shown in Table 4. The enzyme was strongly inhibited by aminopeptidase inhibitors such as actinonin or amastatin.

Table 4

Effect of Each Protease Inhibitor on DLAP1		
Inhibitor	Concentration	Remaining Activity (Relative Value %)
-	-	100
actinonin	100 $\mu$ M	48
amastatin	100 $\mu$ M	30
antipain	100 $\mu$ M	100
alphamenine A	100 $\mu$ M	100
diprotin A	100 $\mu$ M	100
E-64	10 $\mu$ M	80
leuhistin	100 $\mu$ M	83
PMSF	10 mM	100
NEM	100 mM	95
iodoacetamide	50 mM	100
orthophenanthroline	10 $\mu$ M	100

#### (Example 18. Substrate Specificity of DLAP1 for Amino Acid Paranitroanilide Derivatives)

To examine the substrate specificity of DLAP1, the substrate specificity for each amino acid paranitroanilide derivative was compared with that for alanine paranitroanilide (Leu-pNA).

0.025 ml of 2 mM each amino acid paranitroanilide, 0.01 ml of 500 mM HEPES buffer (pH 8.0), and 0.04 ml H<sub>2</sub>O were mixed, then pre-incubated for 5 minutes and reacted with 0.025 ml enzyme solution at 37 °C for 20 minutes. The

reaction was stopped by adding 0.025 ml of 50 % aqueous acetic acid, and the amount of paranitroaniline released in the reaction solution was determined. The enzyme activity causing release of 1  $\mu$ mole paranitroaniline per minute from each amino acid paranitroanilide derivative was assumed to be 1 Unit. The decomposition activity for each substrate, relative to that for alanine paranitroanilide, is shown in Table 5.

5

10

15

20

25

30

Dipeptide	Relative Decomposition Activity
Ala-pNA	100
Gly-pNA	97
Leu-pNA	65
Ile-pNA	21
Val-pNA	23
Phe-pNA	53
Glu-pNA	15
Asp-pNA	10
Lys-pNA	93
Arg-pNA	78
His-pNA	12
Met-pNA	103
(Glu-pNA	9
Pro-pNA	18

The enzymatic properties of DLAP2 obtained in Example 14 in the same manner as DLAP1 are described below. The enzyme activity of DLAP2 was determined by use of L-phenylalanine paranitroanilide in place of the substrate L-leucine paranitroanilide in the method of Example 12.

35

(Example 19. Determination of Optimum pH and Optimum Temperature of DLAP2)

A change in enzyme activity due to reaction pH (optimum pH) was determined in the following manner.

The enzyme reaction buffers used were sodium citrate buffer (pH 3.5, 4.0, 4.5, 5.0, 5.5, 6.0), MES buffer (pH 5.5, 6.0, 6.5, 7.0), HEPES buffer (pH 7.0, 7.5, 8.0), Tris-HCl buffer (pH 8.0, 8.5, 9.0), and sodium carbonate buffer (pH 9.0, 9.5, 10.0, 10.5).

180  $\mu$ l of 0.7 U DLAP2 was prepared in 50 mM buffer at each pH and pre-incubated for 5 minutes at 30 °C. 20  $\mu$ l of the substrate L-phenylalanine paranitroanilide (0.4 mM for reaction) was added to each sample. It was stirred and then incubated for 20 minutes. 50  $\mu$ l of 1 M sodium acetate buffer (pH 4.0) was added to stop the reaction.

The measurement results are shown in FIG. 17. As can be seen from this graph, the optimum pH of DLAP2 of the present invention is in the range of about 5.0 to 9.0, strictly about 6 to about 9 (see FIG. 17).

A change in enzyme activity due to reaction temperature (optimum temperature) was determined as follows:

20  $\mu$ l of the substrate L-phenylalanine paranitroanilide (0.4 mM for reaction) was added to 50 mM MES buffer (pH 6.0) and 180  $\mu$ l of thus obtained solution was pre-incubated for 5 minutes at each temperature (25, 30, 37, 42, 50, 60, 70 °C). 0.7 U DLAP2 was further added to it. The sample was stirred and then incubated for 20 minutes. 50  $\mu$ l of 1 M sodium acetate buffer (pH 4.0) was added to stop the reaction.

The measurement results are shown in FIG. 18. As can be seen from this graph, the optimum temperature of the DLAP2 of the present invention is in the range of about 30 to 70 °C, strictly about 35 to 65 °C (see FIG. 18).

(Example 20. Effect of Protease Inhibitors on DLAP2)

The effect of inhibitors on the DLAP2 of the present invention was examined. The inhibitors used were actinonin,

amastatin, antipain, alphamenine A, diprotin A, leuhistin, phenylmethane sulfonyl fluoride (PMSF), trans-epoxysuccinyl-L-leucylamide (4-guanidino)-butane (E-64), 1,10-orthophenanthroline, N-ethylmaleimide (NEM), and iodoacetamide. The 1,10-orthophenanthroline and N-ethylmaleimide were purchased from Nakarai Tesque K.K. and all the other inhibitors from Peptide Kenkyusho K.K.

In the presence of each inhibitor, the enzyme was left at room temperature (25°C) for 20 minutes, and the remaining activity of the enzyme was determined in the same manner as in Example 12.

The results of activity measurements in the presence of each inhibitor are shown in Table 6. The enzyme was strongly inhibited by actinonin of aminopeptidase inhibitors, and weakly inhibited by PMSF.

Table 6

Effect of Protease Inhibitors on DLAP2		
Inhibitor	Concentration	Remaining Activity (Relative Value %)
-	-	100
actinonin	100 $\mu$ M	37
amastatin	100 $\mu$ M	86
antipain	100 $\mu$ M	100
alphamenine A	100 $\mu$ M	100
diprotin A	100 $\mu$ M	100
E-64	10 $\mu$ M	98
leuhistin	100 $\mu$ M	100
PMSF	10 mM	67
NEM	100 mM	89
iodoacetamide	50 mM	100
orthophenanthroline	10 $\mu$ M	75

#### (Example 21. Substrate Specificity of DLAP2 for Amino Acid Paranitroanilide Derivatives)

To examine the substrate specificity of DLAP2, the substrate specificity for each amino acid paranitroanilide derivative was compared with that for phenylalanine paranitroanilide (Phe-pNA). 0.025 ml of 2 mM each amino acid paranitroanilide, 0.01 ml of 500 mM HEPES buffer (pH 8.0), and 0.04 ml H<sub>2</sub>O were mixed and then reacted with 0.025 ml enzyme solution at 37 °C for 20 minutes. The reaction was stopped by adding 0.025 ml of 50 % aqueous acetic acid, and the amount of paranitroaniline released in the reaction solution was determined. The enzyme activity causing release of 1  $\mu$ mole paranitroaniline per minute from each amino acid paranitroanilide derivative as the substrate was assumed to be 1 Unit. The decomposition activity for each substrate, relative to that for phenylalanine paranitroanilide, is shown in Table 7.



Table 7

Synthetic Substrate	Relative Decomposition Activity (%)
Phe-pNA	100
Ala-pNA	3
Gly-pNA	2
Leu-pNA	5
Ile-pNA	0
Val-pNA	0
Glu-pNA	0
Asp-pNA	0
Lys-pNA	0
Arg-pNA	0
His-pNA	0
Met-pNA	3
Glu-pNA	0
Pro-pNA	3

(Example 22. Decomposition of Soybean Protein with Aminopeptidase GX, D3, DLAP1, and DLAP2)

A method of decomposing soybean protein with the above characterized aminopeptidase GX, D3, DLAP1 and DLAP2 prepared from germinated soybean cotyledons in the manner described above is described below.

#### 1. Primary Decomposition of Soybean Protein with Protease D3

The soybean protein used was a soybean protein preparation, Ajipron SU (Ajinomoto Co. Inc). An aqueous solution of Ajipron SU (2.5 g/35 ml Milli-Q water) was sterilized in an autoclave at 120 °C for 10 minutes and then adjusted to pH 4.0 with HCl. This substrate solution (2.5 g/40 ml) was reacted with the refolded pD3- $\beta$  from the recombinant *E. coli* (FERM BP-5793) prepared in Example 11. For this reaction, 1 ml of the refolded pD3- $\beta$  (25 U/ml), 0.4 ml of 5 M aqueous sodium chloride and 0.5 ml of 20 mM aqueous cysteine were mixed with 8 ml of the above substrate solution and reacted at 37 °C for 0 to 72 hours. The progress of the decomposition was monitored by quantifying nitrogen in the form of amino and imino as described in Japanese Patent LOP Publication No. 264/1996 with 4-nitro-7-nitrobenzo-2-oxa-1,3-diazole (NBD-F) reagent (K. Imai, and Y. Watanabe, Anal. Chim. Acta., 130, 377-383 (1983)). The results are shown in Table 8.

Table 8

Reaction Time	NBD-F value (mM)
0 hour	N.D.
24 hours	60
48 hours	89
72 hours	106

## 2. Secondary Decomposition of Soybean Protein with Aminopeptidase GX

The primary hydrolysate obtained after 72-reaction in 1 above was decomposed with GX prepared in Example 3. The primary hydrolysate obtained in 1 above was adjusted to about pH 7 with sodium hydroxide. 200  $\mu$ l of 1 U GX was added to 200  $\mu$ l of the primary hydrolysate (containing 10 mg substrate) and 400  $\mu$ l of thus obtained solution was reacted at 42°C for 72 hours (Group 1). 200  $\mu$ l of 1 U GX was made 400  $\mu$ l(not containing the substrate) and incubated at 42 °C for 72 hours (Control 1), and 200 $\mu$ l of the primary hydrolysate (containing 10 mg substrate) obtained in 1 above was made 400  $\mu$ l and incubated at 42 °C for 72 hours (Control 2).

The amino acid analysis of the hydrolysates after 72-hour reaction was carried out. The results are shown in Table 9.

As shown in Table 9, both glutamic acid and aspartic acid in group 1 were released much more than other amino acids.

Table 9

Amino Acid Analysis in Each Group After 72 Hours (mM)			
	Group 1	Control 1	Control 2
aspartic acid	2.6	N.D.	0.3
asparagine	1.5	N.D.	0.4
glutamic acid	5.4	N.D.	1.6
glutamine	0.8	N.D.	0.6
threonine	0.8	N.D.	0.5
serine	2.7	N.D.	0.6
proline	0	N.D.	0
glycine	1.7	N.D.	0.5
alanine	2.4	N.D.	0.8
valine	0.9	N.D.	0.9
cystine	2.0	N.D.	2.0
methionine	0.7	N.D.	0.7
isoleucine	1.4	N.D.	1.2
leucine	1.7	N.D.	1.5
tyrosine	0	N.D.	0
phenylalanine	0.6	N.D.	0.5
tryptophan	0	N.D.	0
lysine	1.0	N.D.	0.8
histidine	0.4	N.D.	0.3
arginine	1.1	N.D.	0.9
amino acids in total	27.1	N.D.	13.9

## 3. Secondary Decomposition of Soybean Protein with GX, DLAP1 and DLAP2

The primary hydrolysate obtained after 72-hour reaction in 1 above was decomposed with aminopeptidase GX and DLAP1 and DLAP2 prepared in Example 14. The primary hydrolysate obtained in 1 above was adjusted to about pH 7 with sodium hydroxide. 350  $\mu$ l of a mixture of GX (1 U), DLAP1 (320 U) and DLAP2 (100 U) was added to 50  $\mu$ l of the primary hydrolysate (containing 2.5 mg substrate) and 400  $\mu$ l of thus obtained solution was reacted at 42°C for 0 to 72 hours (Group 2). Separately, 350  $\mu$ l of a mixture of GX (1 U), DLAP1 (320 U) and DLAP2 (100 U) was made 400  $\mu$ l and

incubated at 42 °C for 0 to 72 hours (Control 3). Only 50 µl of the primary hydrolysate (containing 2.5 mg substrate) obtained in 1 above was made 400 µl and incubated at 42 °C for 0 to 72 hours (Control 4). The hydrolysate after 72-hour reaction in Group 2 was further hydrolyzed with constant-boiling HCl (Wako Pure Chemical Industries) in a vacuum ample at 110 °C for 24 hours. The progress of the decomposition was monitored using NBD-F in the same manner as in above 1. The results are shown in Tabl 10.

Table 10

Reaction Time	NBD-F value (mM)		
	Group 2	Control 3	Control 4
0 hour	13	8	26
24 hours	35	9	25
48 hours	39	9	27
72 hours	38	9	25

The amino acid analysis of the hydrolysates after 72-hour reaction was carried out. The amino acid contents in the acid hydrolysate of Group 2 was expressed relatively to those in the secondary hydrolysate of Group 2 (Table 11).

Table 11

Amino Acid Analysis in Each Group After 72-Hour Reaction (mM)				
	Group 2	Acid Hydrolysate of Group 2	Group 3	Group 4
aspartic acid	1.3	3.9	0.1	0.2
asparagine	1.4	0	0	0.2
glutamic acid	2.5	6.2	0.2	0.8
glutamine	1.5	0	0	0.3
threonine	1.3	1.4	0.2	0.3
serine	2.4	2.3	0.1	0.3
proline	1.8	2.1	0.2	0
glycine	2.4	2.4	0.2	0.2
alanine	2.3	1.9	0.3	0.4
valine	2.1	1.6	0.3	0.3
cystine	0.7	0.6	0	1.0
methionine	0.6	0.4	0.4	0.4
isoleucine	1.7	1.4	0.3	0.6
leucine	3.0	2.5	0.5	0.7
tyrosine	0.9	0.9	0.1	0
phenylalanine	1.3	1.3	0.1	0.3
tryptophan	0	0	0	0
lysine	2.1	1.9	0.2	0.4
histidine	0.7	0.7	0.1	0.2
arginine	2.1	2.0	0.1	0.4
amino-acids in total	32.1	33.5	3.4	7.0

As shown in Table 11, the total amino acids in the secondary hydrolysate of Group 2 are 32.1 mM and the total amino acids in the acid hydrolysate of Group 2 are 33.5 mM. In Group 2, the yield of the amino acids in the secondary hydrolysate Group 2 is therefore about 95.8 % of the acid hydrolysate.

In group 2, the fact that other amino acids besides both aspartic acid and glutamic acid were released highly would be effect of both DLAP 1 and DLAP 2 in addition to that of amino peptidase GX.

#### [Effect of the Invention]

The novel aminopeptidase GX of the present invention is considered to be an enzyme which is extremely useful in producing highly decomposed hydrolysates with high contents of acidic amino acids from starting materials containing peptides and proteins. That is, as shown in Example 8, the enzyme of the present invention highly releases acidic amino acids which are hardly released with even aminopeptidase M having broad substrate specificity.

The novel aminopeptidase GX of the present invention can be used in combination with the conventional protein decomposition method, so that those substrates particularly having N-terminal acidic amino acid still not released by the conventional method can be decomposed. More specifically, with the aminopeptidase GX of the present invention given, a peptide or protein containing glutamic acid or aspartic acid at the N-terminal can release the glutamic acid or aspartic acid participating in good tastes. The peptide or protein hydrolysate thus obtained can be used to various materials such as foods, seasonings, feeds etc. Particularly, as shown in Example 22, protease D3, DLAP1 and DLAP2 which all were derived from germinated soybean cotyledons can be used in combination to decompose soybean protein and a combination of these enzymes results in about 95.8 % yield based on a hydrochloric acid hydrolysate of soybean protein.

In addition, major components in soybean protein, e.g.  $\beta$ -conglisinin, glycine etc., are known to contain acidic amino acid-enriched sequences such as -Glu-Glu-Glu-Glu-Glu- etc. (J. J. Doyle et al., J. Biol. Chem., 261 9228 (1986), J. D. Ng et al., Plant Physiol., 101, 713 (1993), N. C. Nielsen et al., Plant Cell 1 313 (1989)), and these sequences would not be decomposed with the conventional protease and peptidase. The aminopeptidase GX of the present invention derived from germinated soybean cotyledons is considered to have best specificity and properties as an enzyme for efficiently decomposing acidic peptides derived from soybean protein. It is therefore believed that the enzyme of the present invention, aminopeptidase GX, is the most effective aminopeptidase for improving the release of acidic amino acids.

Although the aminopeptidase specific for acidic amino acids includes those derived from porcine digestive tract membrane (A. Benajiba et al., Eur. J. Biochem. 107, 381 (1980)) and from lactic acid bacteria (F. A. Exterkate et al., Appln. Environ. Microbiol., 53, 577 (1987)) as EC. 3.4.11.7 glutamylaminopeptitase, the aminopeptidase GX of the present invention is not inhibited by an inhibitor (amastatin) of such conventional aminopeptidase. Further, the aminopeptidase GX of the present invention is inhibited by leuhistine and actinonin, but no report has been made of glutamylaminopeptidase inhibited by said inhibitors. An aminopeptidase derived from yeast of the genus *Rhodotolura* (Japanese Patent LOP Publication No. 244,381/1987) is also an acidic amino acids specific aminopeptidase, but this enzyme is different from the enzyme of the present invention because it is not inhibited by magnesium chloride or manganese chloride. Another aminopeptidase from soybean is also reported (Shinji Watanabe et al., Nippon Nogei Kagakkaishi 63(3), 617 (1988)), but this prior enzyme completely differs in substrate specificity from the aminopeptidase GX of the present invention. Further, ever reported aminopeptidases derived from other plants include none of an enzyme similar to the enzyme of the present invention, and it can thus be concluded that the enzyme of the present invention is an acidic amino acid-specific aminopeptidase found for the first time as being derived from a plant. Therefore, the novel aminopeptidase GX of the present invention is remarkable from a biological viewpoint as well.

SEQUENCE LISTING

5

(1) GENERAL INFORMATION:

10

(i) APPLICANT:

15

- (A) NAME: Ajinomoto Co., Inc.
- (B) STREET: No. 15-1, Kyobashi 1-chome
- (C) CITY: Chuo-ku, Tokyo
- (E) COUNTRY: Japan
- (F) POSTAL CODE (ZIP): None

20

- (ii) TITLE OF INVENTION: Aminopeptidase GX, and a method of  
hydrolyzing a protein with the same

- (iii) NUMBER OF SEQUENCES: 2

25

(iv) COMPUTER READABLE FORM:

30

- (A) MEDIUM TYPE: Floppy disk
- (B) COMPUTER: IBM PC compatible
- (C) OPERATING SYSTEM: PC-DOS/MS-DOS
- (D) SOFTWARE: PatentIn Release #1.0, Version #1.30 (EPO)

35

(v) CURRENT APPLICATION DATA:

APPLICATION NUMBER: EP 97 103 862.5

(2) INFORMATION FOR SEQ ID NO: 1:

40

(i) SEQUENCE CHARACTERISTICS:

45

- (A) LENGTH: 1056 base pairs
- (B) TYPE: nucleic acid
- (C) STRANDEDNESS: double
- (D) TOPOLOGY: linear

50

- (ii) MOLECULE TYPE: cDNA to mRNA

55

- (vi) ORIGINAL SOURCE:

EP 0 794 253 A2

(A) ORGANISM: Glycine max

5

(ix) FEATURE:

(A) NAME/KEY: CDS

(B) LOCATION:1..1056

10

(xi) SEQUENCE DESCRIPTION: SEQ ID NO: 1:

15	ATG GCT AGG TAC GAC AGC GCC CAC GCG GAC AAG GCC GCC ACG TTG CGC	48
	Met Ala Arg Tyr Asp Ser Ala His Ala Asp Lys Ala Ala Thr Leu Arg	
	1 5 10 15	
20	ACC GAG GAG GAG CTG ATG TCC ATG TAC GAG CAG TGG CTC GTG AAG CAC	96
	Thr Glu Glu Glu Leu Met Ser Met Tyr Glu Gln Trp Leu Val Lys His	
	20 25 30	
25	GGG AAG GTG TAC AAC GCG CTC GGC GAG AAG GAG AAG CGC TTC CAG ATC	144
	Gly Lys Val Tyr Asn Ala Leu Gly Glu Lys Glu Lys Arg Phe Gln Ile	
	35 40 45	
30	TTC AAG GAC AAC CTG CGA TTC ATC GAC GAC CAC AAC TCC GCG GAG GAC	192
	Phe Lys Asp Asn Leu Arg Phe Ile Asp Asp His Asn Ser Ala Glu Asp	
35	50 55 60	
	CGA ACC TAC AAG CTC GGA CTG AAC CGG TTC GCT GAT CTC ACC AAC GAG	240
40	Arg Thr Tyr Lys Leu Gly Leu Asn Arg Phe Ala Asp Leu Thr Asn Glu	
	65 70 75 80	
	GAA TAC AGG GCC AAG TAC TTG GGA ACC AAG ATC GAT CCC AAC CGG AGG	288
45	Glu Tyr Arg Ala Lys Tyr Leu Gly Thr Lys Ile Asp Pro Asn Arg Arg	
	85 90 95	
50	CTC GGA AAG ACC CCG AGC AAC CGC TAC GCG CCA CGT GTC GGC GAC AAA	336
	Leu Gly Lys Thr Pro Ser Asn Arg Tyr Ala Pro Arg Val Gly Asp Lys	
	100 105 110	

55

EP 0 794 253 A2

5	TTG CCT GAT TCC GTT GAT TGG AGG AAG GAA GGT GCT GTT CCT CCT GTC Leu Pro Asp Ser Val Asp Trp Arg Lys Glu Gly Ala Val Pro Pro Val	384
	115 120 125	
10	AAA GAC CAA GGA GGC TGT GGG AGC TGT TGG GCA TTC TCA GCA ATC GGT Lys Asp Gln Gly Gly Cys Gly Ser Cys Trp Ala Phe Ser Ala Ile Gly	432
	130 135 140	
15	GCA GTA GAA GGA ATA AAT AAG ATA GTA ACA GGC GAA CTG ATT TCG TTA Ala Val Glu Gly Ile Asn Lys Ile Val Thr Gly Glu Leu Ile Ser Leu	480
	145 150 155 160	
20	TCA GAA CAA GAA TTG GTG GAT TGT GAT ACT GGA TAT AAC CAA GGA TGC Ser Glu Gln Glu Leu Val Asp Cys Asp Thr Gly Tyr Asn Gln Gly Cys	528
	165 170 175	
25	AAT GGA GGA CTT ATG GAC TAT GCA TTT GAG TTC ATA ATC AAC AAT GGC Asn Gly Gly Leu Met Asp Tyr Ala Phe Glu Phe Ile Ile Asn Asn Gly	576
	180 185 190	
30	GGC ATT GAT TCT GAT GAG GAT TAC CCA TAC CGT GGT GTT GAT GGT AGA Gly Ile Asp Ser Asp Glu Asp Tyr Pro Tyr Arg Gly Val Asp Gly Arg	624
	195 200 205	
35	TGC GAC ACA TAT AGG AAA AAT GCT AAA GTC GTT TCT ATT GAT GAC TAC Cys Asp Thr Tyr Arg Lys Asn Ala Lys Val Val Ser Ile Asp Asp Tyr	672
	210 215 220	
40	GAA GAT GTT CCT GCC TAT GAT GAG TTA GCC TTG AAA AAG GCC GTT GCA Glu Asp Val Pro Ala Tyr Asp Glu Leu Ala Leu Lys Lys Ala Val Ala	720
	225 230 235 240	
45	AAT CAG CCC GTG AGC GTT GCT ATT GAA GGA GGG GGC AGG GAA TTT CAA Asn Gln Pro Val Ser Val Ala Ile Glu Gly Gly Gly Arg Glu Phe Gln	768
	245 250 255	
50	TTA TAT GTA TCT GGT GTA TTC ACG GGG AGA TGT GGC ACA GCA CTA GAT	816
55		

EP 0 794 253 A2

Leu Tyr Val Ser Gly Val Phe Thr Gly Arg Cys Gly Thr Ala Leu Asp  
260 265 270

5

CAT GGT GTC GTG GCT GTT GGG TAT GGA ACA GCT AAA GGT CAT GAT TAT 864  
His Gly Val Val Ala Val Gly Tyr Gly Thr Ala Lys Gly His Asp Tyr  
275 280 285

10

TGG ATC GTA AGG AAT TCA TGG GGT TCT AGC TGG GGA GAG GAT GGC TAC 912  
Trp Ile Val Arg Asn Ser Trp Gly Ser Ser Trp Gly Glu Asp Gly Tyr  
290 295 300

15

ATC AGA TTA GAA AGA AAT CTT GCT AAC AGC AGA TCA GGC AAG TGT GGA 960  
Ile Arg Leu Glu Arg Asn Leu Ala Asn Ser Arg Ser Gly Lys Cys Gly  
305 310 315 320

20

ATT GCA ATT GAG CCA TCT TAT CCC CTT AAG AAT GGT CCA AAT CCC CCT 1008  
Ile Ala Ile Glu Pro Ser Tyr Pro Leu Lys Asn Gly Pro Asn Pro Pro  
325 330 335

25

AAT CCT GGA CCA TCA CCC CCT TCA CCT GTG AAG CCG CCA AAT GTC TGA 1056  
Asn Pro Gly Pro Ser Pro Pro Ser Pro Val Lys Pro Pro Asn Val \*  
340 345 350

30

35

(2) INFORMATION FOR SEQ ID NO: 2:

(i) SEQUENCE CHARACTERISTICS:

40

- (A) LENGTH: 352 amino acids
- (B) TYPE: amino acid
- (D) TOPOLOGY: linear

45

(ii) MOLECULE TYPE: protein

(xi) SEQUENCE DESCRIPTION: SEQ ID NO: 2:

50

Met Ala Arg Tyr Asp Ser Ala His Ala Asp Lys Ala Ala Thr Leu Arg  
1 5 10 15

55



# EP 0 794 253 A2

Thr Glu Glu Glu Leu Met Ser Met Tyr Glu Gln Trp Leu Val Lys His  
 20 25 30  
 5  
 Gly Lys Val Tyr Asn Ala Leu Gly Glu Lys Glu Lys Arg Phe Gln Ile  
 35 40 45  
 10  
 Phe Lys Asp Asn Leu Arg Phe Ile Asp Asp His Asn Ser Ala Glu Asp  
 50 55 60  
 15  
 Arg Thr Tyr Lys Leu Gly Leu Asn Arg Phe Ala Asp Leu Thr Asn Glu  
 65 70 75 80  
 20  
 Glu Tyr Arg Ala Lys Tyr Leu Gly Thr Lys Ile Asp Pro Asn Arg Arg  
 85 90 95  
 25  
 Leu Gly Lys Thr Pro Ser Asn Arg Tyr Ala Pro Arg Val Gly Asp Lys  
 100 105 110  
 30  
 Leu Pro Asp Ser Val Asp Trp Arg Lys Glu Gly Ala Val Pro Pro Val  
 115 120 125  
 35  
 Lys Asp Gln Gly Gly Cys Gly Ser Cys Trp Ala Phe Ser Ala Ile Gly  
 130 135 140  
 40  
 Ala Val Glu Gly Ile Asn Lys Ile Val Thr Gly Glu Leu Ile Ser Leu  
 145 150 155 160  
 45  
 Ser Glu Gln Glu Leu Val Asp Cys Asp Thr Gly Tyr Asn Gln Gly Cys  
 165 170 175  
 50  
 Asn Gly Gly Leu Met Asp Tyr Ala Phe Glu Phe Ile Ile Asn Asn Gly  
 180 185 190  
 Gly Ile Asp Ser Asp Glu Asp Tyr Pro Tyr Arg Gly Val Asp Gly Arg  
 195 200 205  
 55  
 Cys Asp Thr Tyr Arg Lys Asn Ala Lys Val Val Ser Ile Asp Asp Tyr

	210	215	220
5	Glu Asp Val Pro Ala Tyr Asp Glu Leu Ala Leu Lys Lys Ala Val Ala		
	225	230	235 240
10	Asn Gln Pro Val Ser Val Ala Ile Glu Gly Gly Gly Arg Glu Phe Gln		
		245	250 255
15	Leu Tyr Val Ser Gly Val Phe Thr Gly Arg Cys Gly Thr Ala Leu Asp		
		260	265 270
20	His Gly Val Val Ala Val Gly Tyr Gly Thr Ala Lys Gly His Asp Tyr		
		275	280 285
25	Trp Ile Val Arg Asn Ser Trp Gly Ser Ser Trp Gly Glu Asp Gly Tyr		
		290	295 300
30	Ile Arg Leu Glu Arg Asn Leu Ala Asn Ser Arg Ser Gly Lys Cys Gly		
		305	310 315 320
35	Ile Ala Ile Glu Pro Ser Tyr Pro Leu Lys Asn Gly Pro Asn Pro Pro		
		325	330 335
40	Asn Pro Gly Pro Ser Pro Pro Ser Pro Val Lys Pro Pro Asn Val *		
		340	345 350

#### 45 Claims

##### 1. Aminopeptidase GX having the following properties:

- |    |   |
|----|---|
|    | 1) optimum pH: about 5.5 to about 9.5;  |
| 50 | 2) optimum temperature: about 25 to about 60 °C;  |
|    | 3) temperature stability: keeping about 80 % or more activity after left at 50 °C for 80 minutes or about 40 % or more activity after left at 60 °C for 40 minutes;   |
|    | 4) molecular weight: about 400 to 550 kD (gel filtration),  |
| 55 | about 380 to 460 kD (native PAGE), and  |
|    | about 53 to 60 kD, about 30 to 32 kD,   |
|    | and about 25 to 28 kD (SDS-PAGE after reduction and heating);   |
|    | 5) substrate specificity: decomposing a peptide or protein containing glutamic acid or aspartic acid at the N-terminal to release the glutamic acid or aspartic acid; |
|    | 6) inhibitors: inhibited by leuhistin, actinonin, alphamenine A or 1,10-orthophenanthroline; and  |

7) effect of metal ions: inhibited by magnesium or copper.

2. The aminopeptidase according to claim 1, which is derived from germinated soybean cotyledons.

5 3. A method of hydrolyzing a peptide or protein, which comprises allowing the aminopeptidase GX of claim 1 or 2 to contact and react with a peptide or protein.

4. A method of hydrolyzing a peptide or protein, which comprises allowing a cell extract obtained by disrupting germinated soybean cotyledons containing the aminopeptidase GX of claim 1 or 2 to contact and react with a peptide or protein.  
10

5. The method according to claim 3 or 4, wherein cysteine protease D3 derived from germinated soybean cotyledons or a crude enzyme solution containing the same is used in combination.

15 6. The method according to claim 5, wherein leucine aminopeptidase or a crude enzyme solution containing the same is further used in combination.

20

25

30

35

40

45

50

55

Fig.1

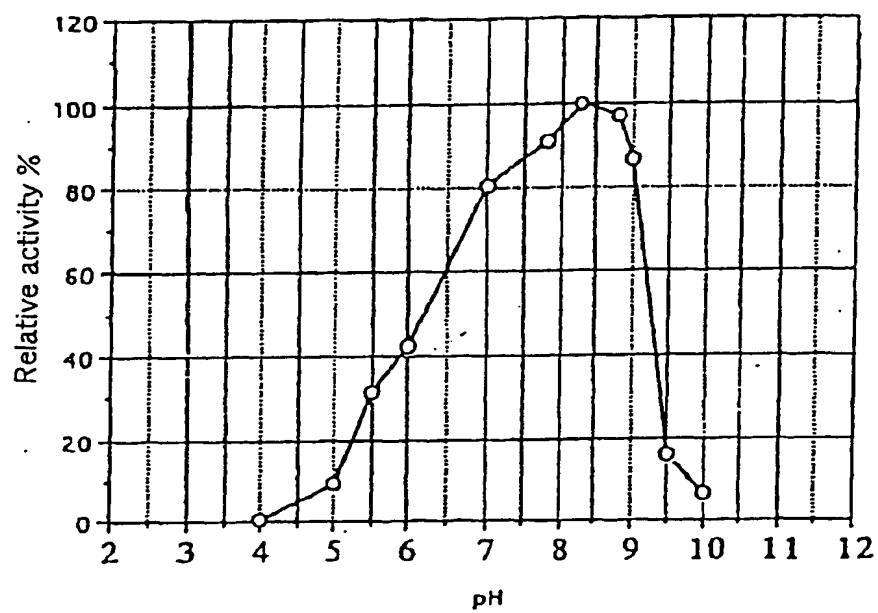


Fig.2

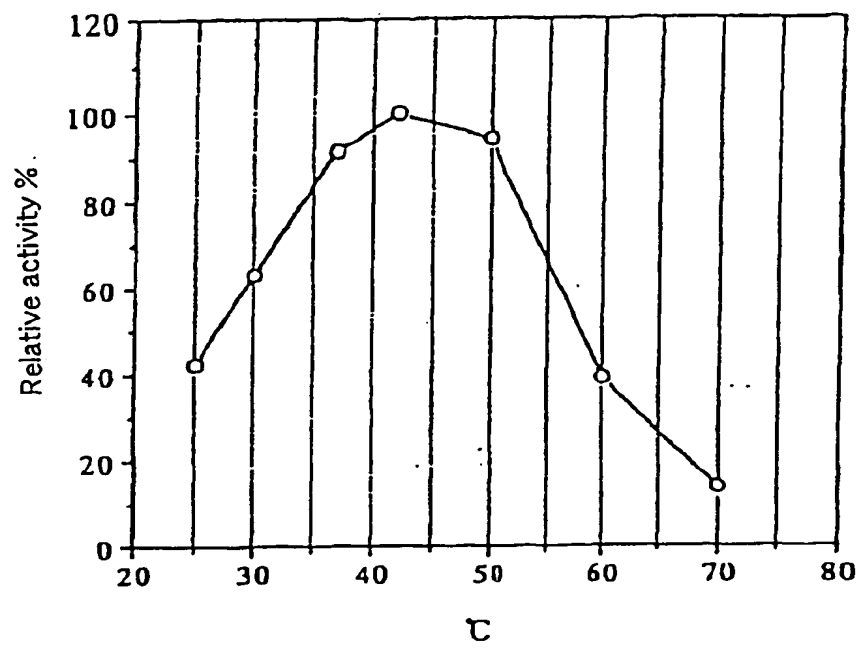


Fig. 3

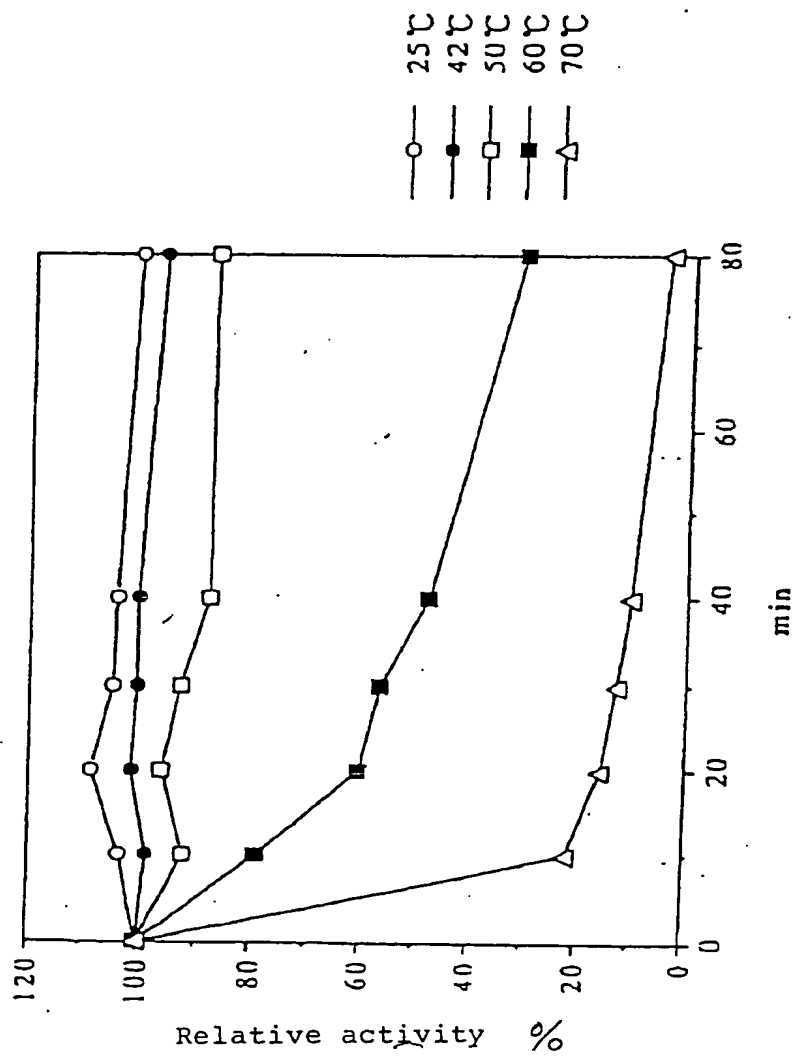
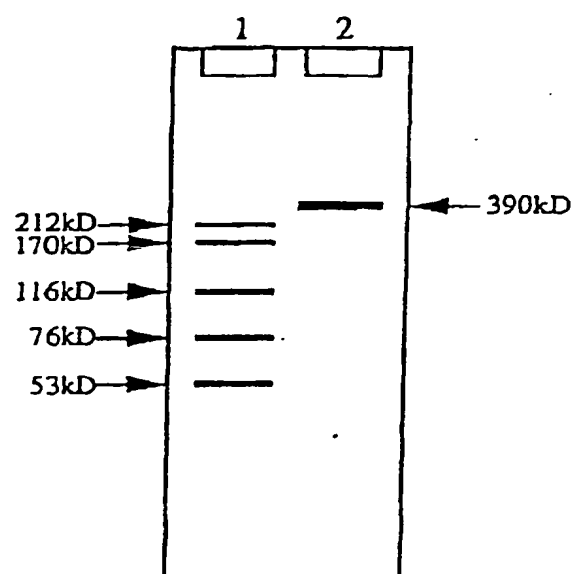


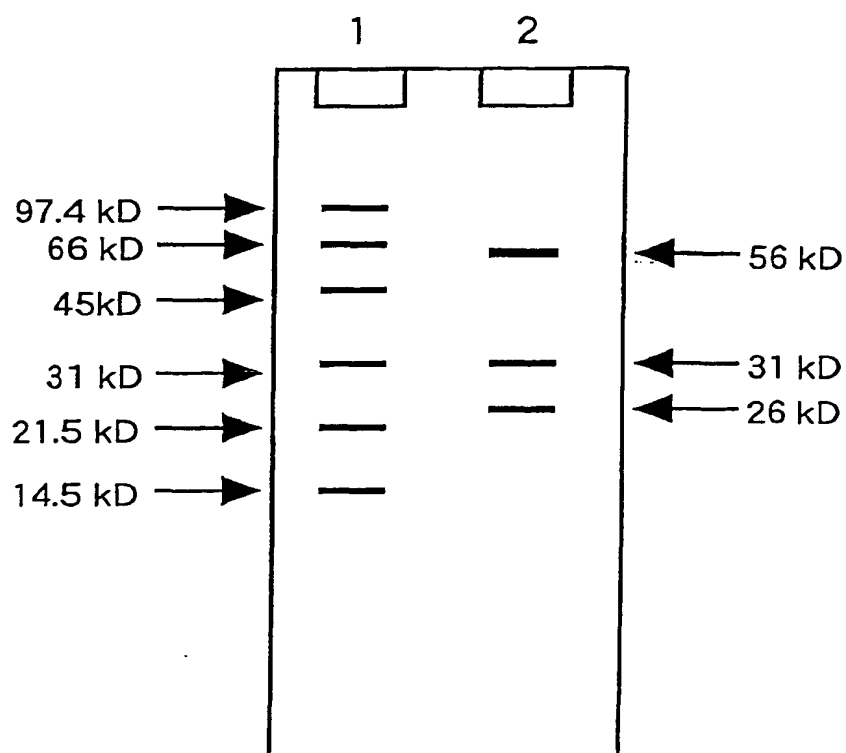
Fig.4



Lane 1 : Molecular weight marker

Lane 2 : 20  $\mu$ g GX

Fig. 5



Lane1: Molecular weight marker

Lane2: 20  $\mu$ g GX



Fig.6

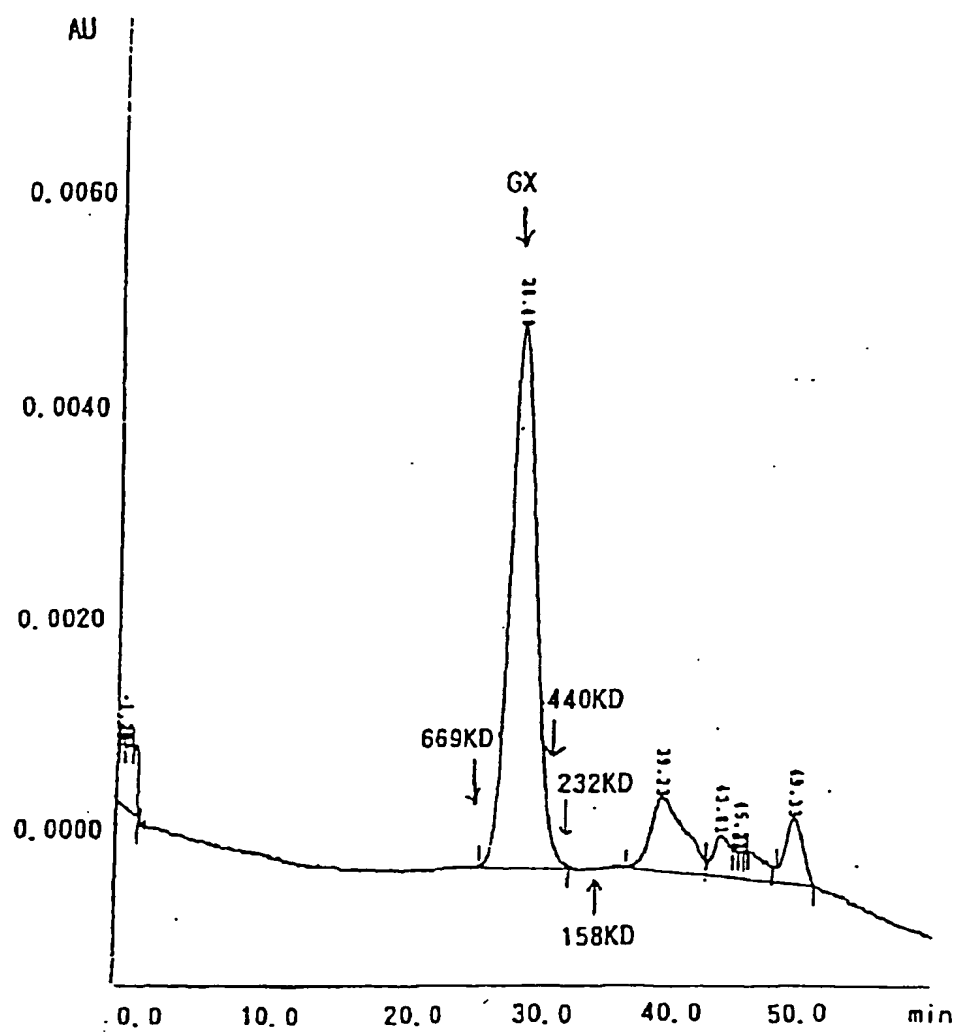


Fig.7

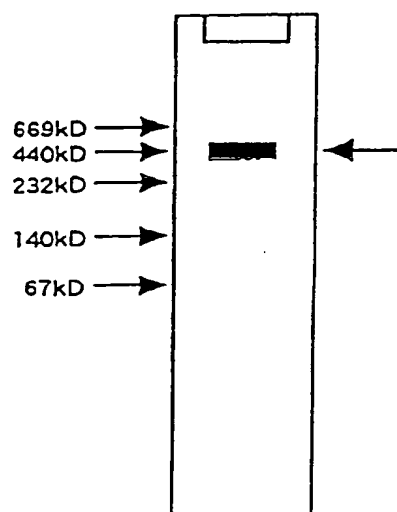


Fig.8

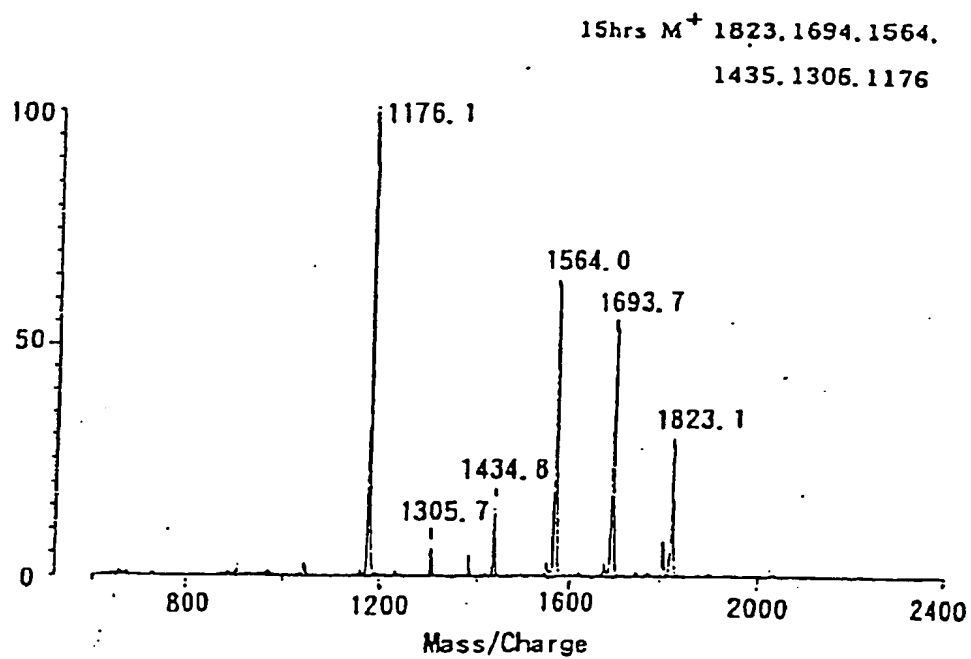
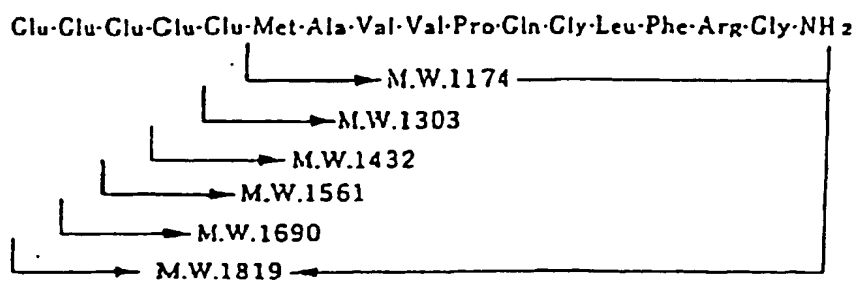


Fig.9

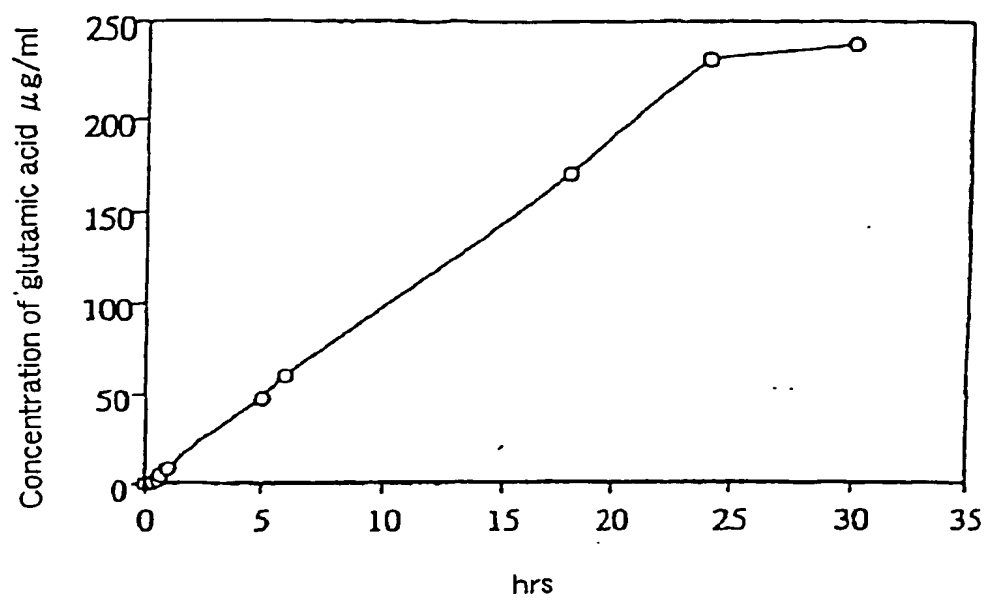


Fig.10

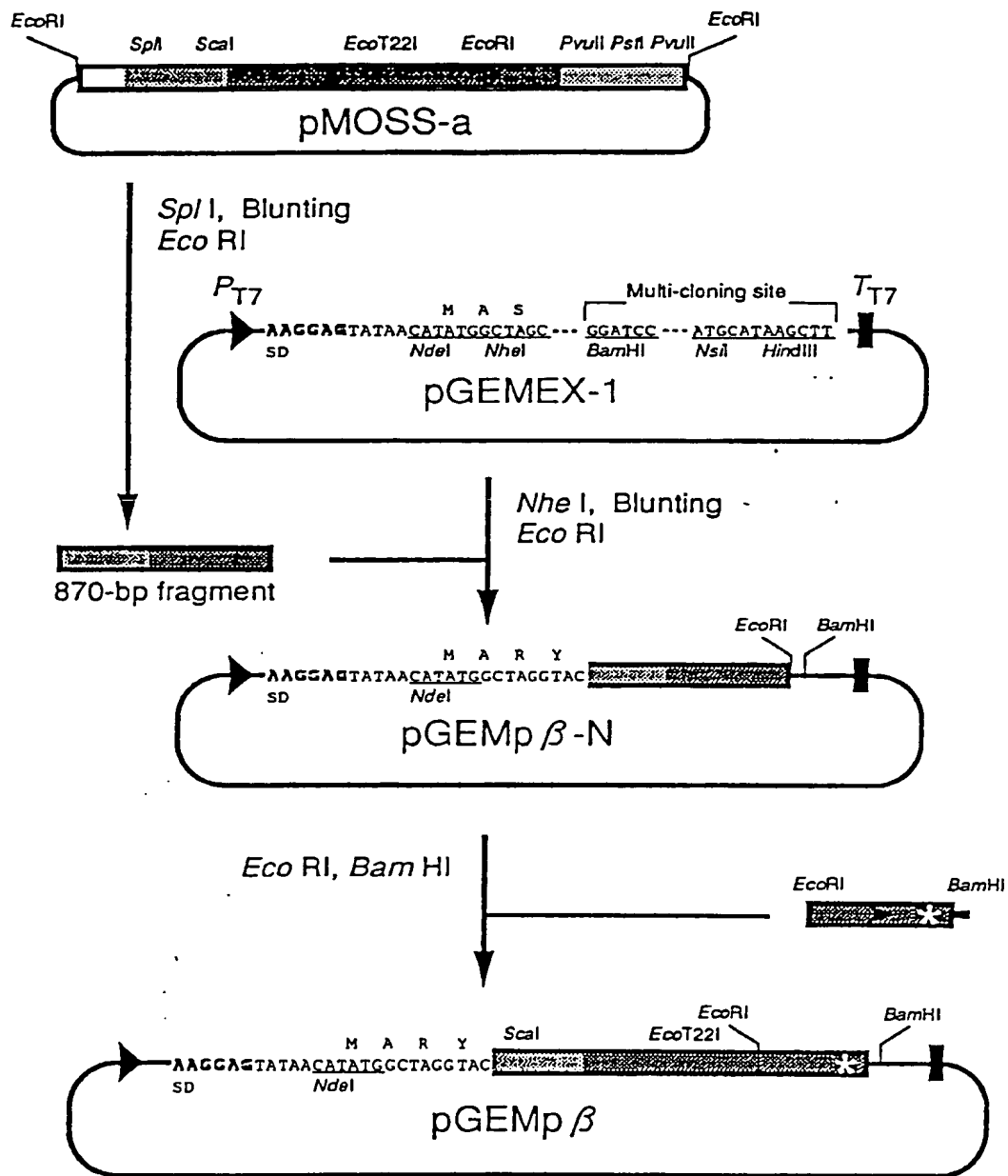


Fig 11

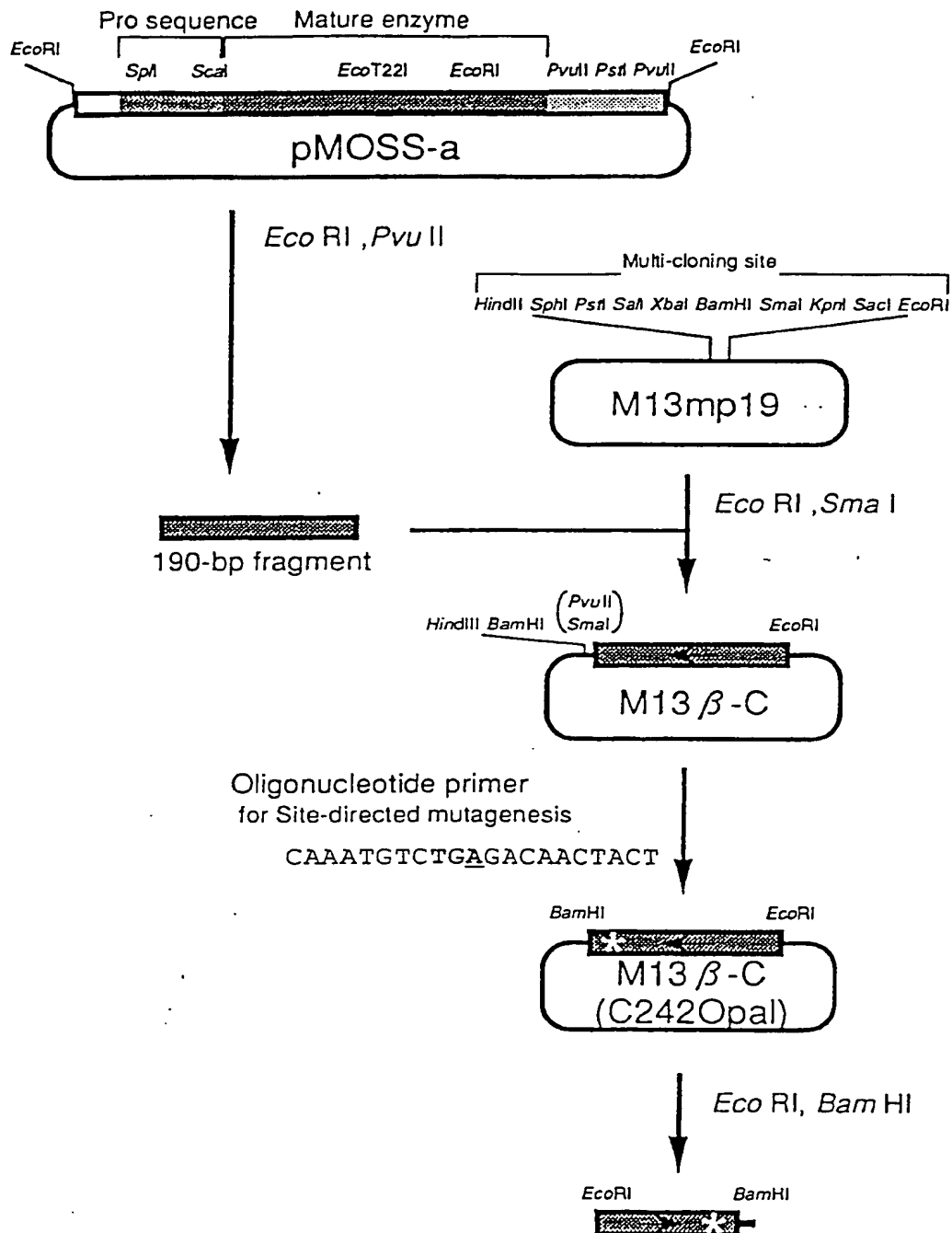
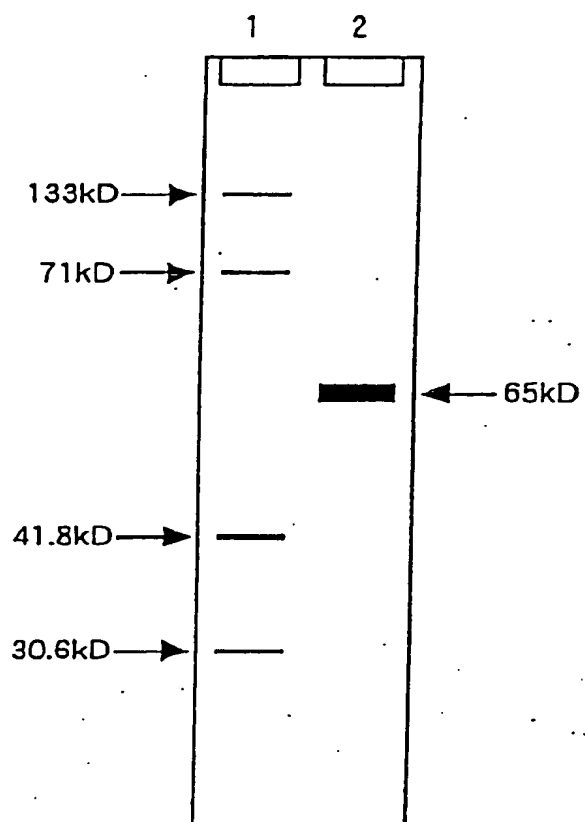


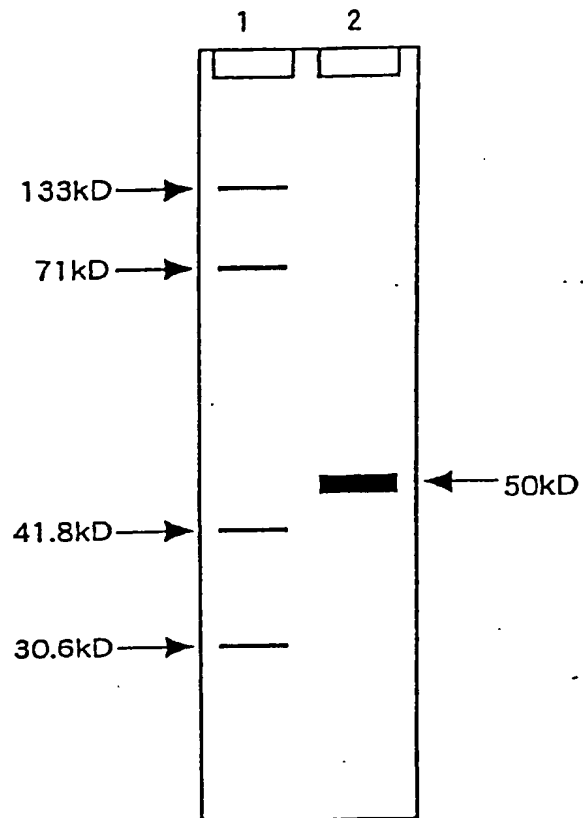
Fig.12



Lane 1 : Molecular weight marker

Lane 2 : DLAP1

Fig.13



Lane 1 : Molecular weight marker  
Lane 2 : DLAP 2



Fig.14

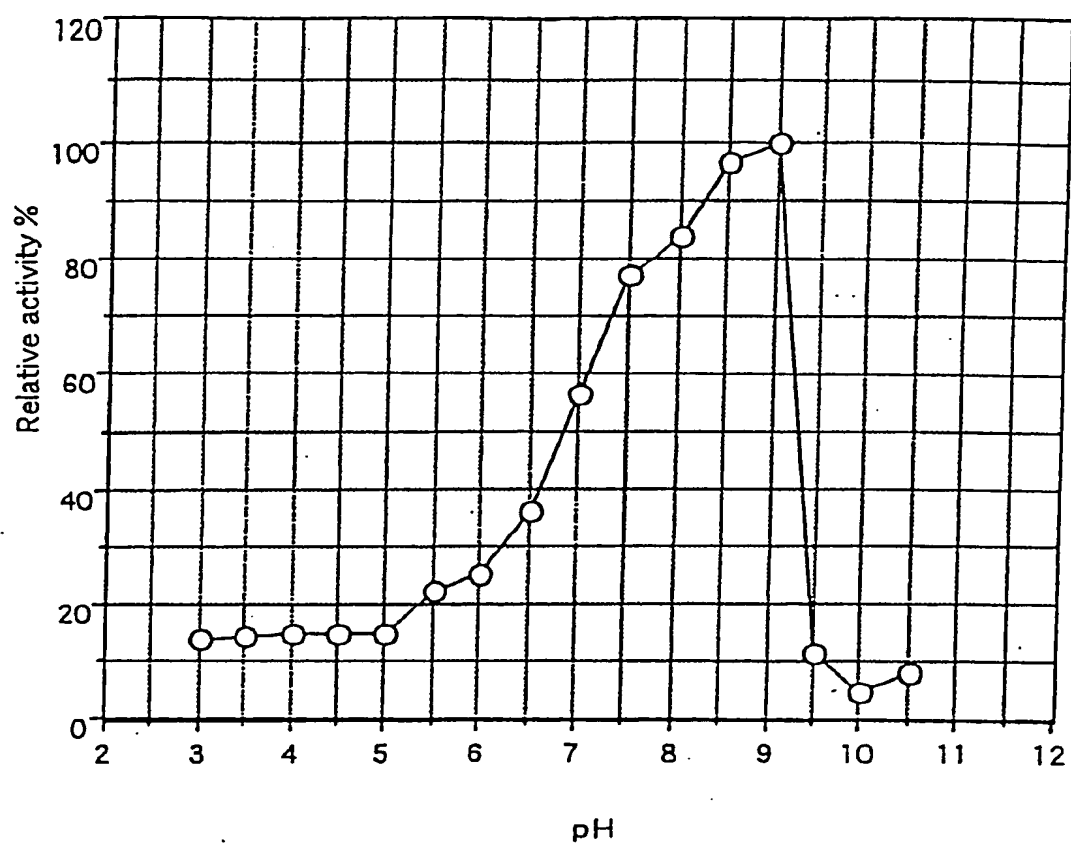


Fig.15

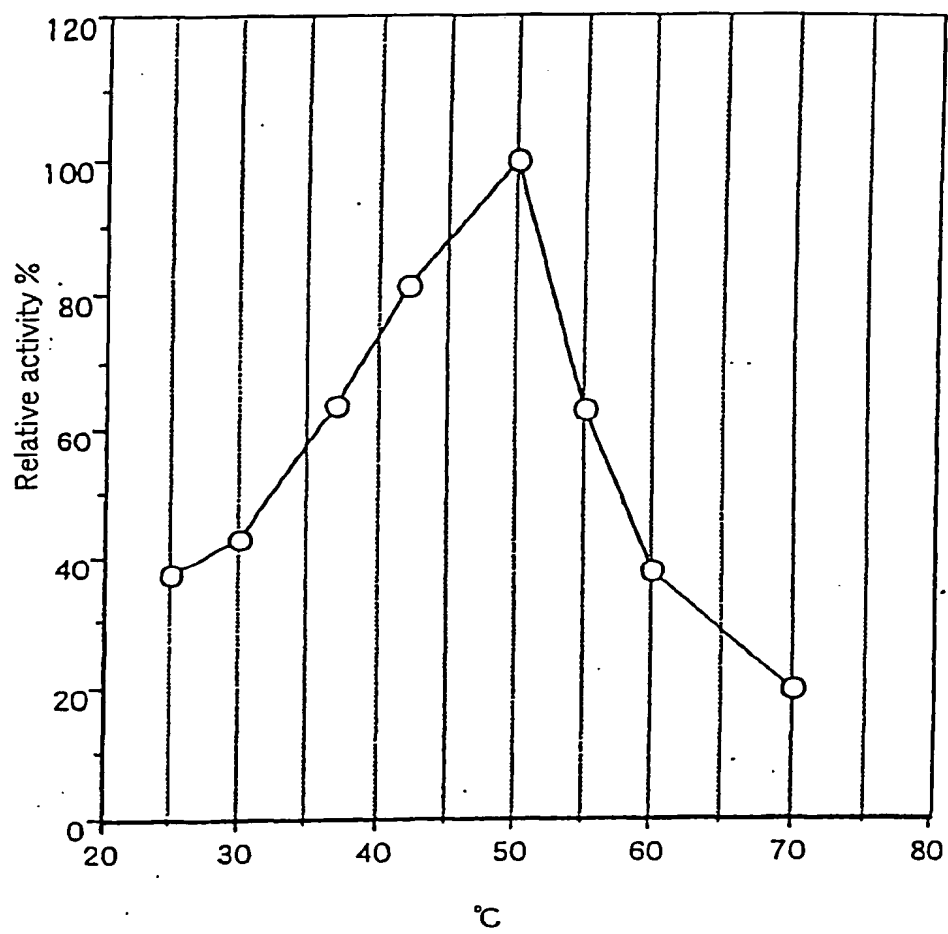


Fig.16

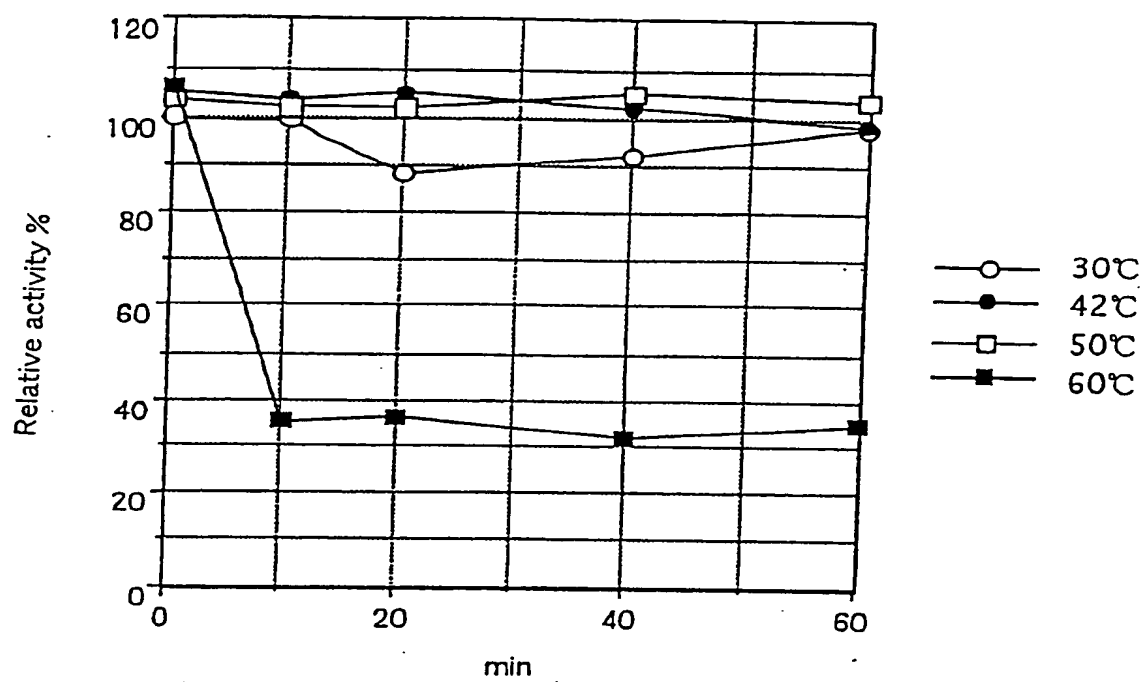


Fig.17

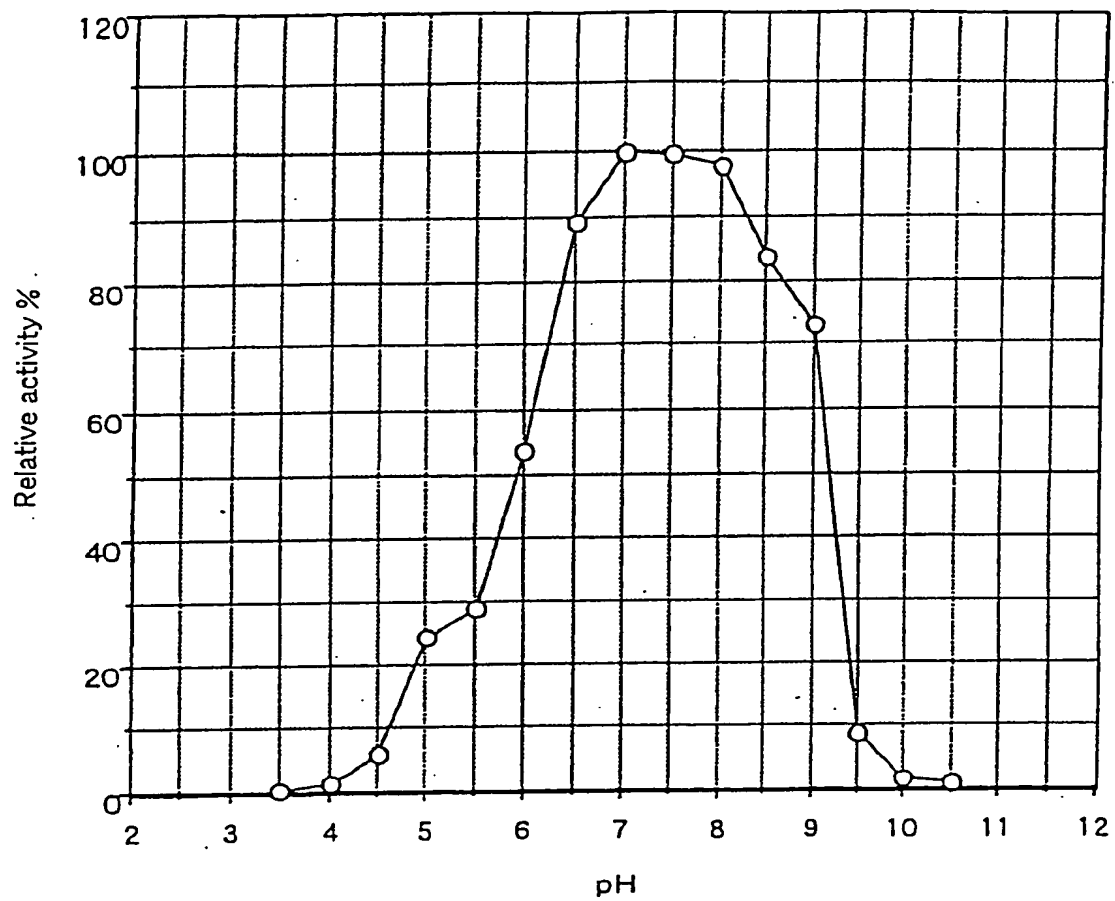


Fig.18

